# TAIWAN Photon Source

National Synchrotron Radiation Research Center

### **Novel Nano materials and applications**

Tseng, Shao-Chin 2018/09/20

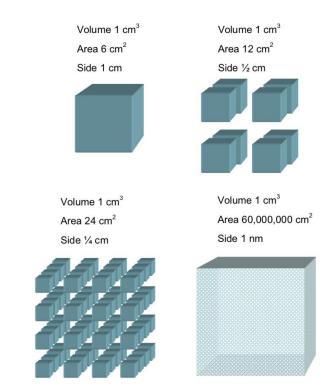
X-ray and IR Image Group Experiment Facility Division NSRRC

www.nsrrc.org.tw

# Outline

- Advantage of materials in nanometer dimensions
- Nanomaterial analysis technology





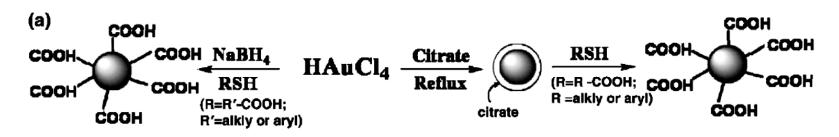
What is the nanotechnology? The term nanotechnology is employed to describe the creation and exploitation of materials with structural features in between those of atoms and bulk materials, with at least dimension in the nanometer range (1 nm = 10-9 m).

There are many uses and applications available using nanotechnology that are not possible using conventional materials which make it unique. For applications that use a substance's chemical properties, substantially less nanomaterial may be required to do the job of a conventional material. The chemical reactivity of a material is related to its surface area compared to its volume and the surface area for a nanoparticle is enormous per unit volume. The diagram below illustrates how surface area increases when a material is dissected into nano-sized particles.

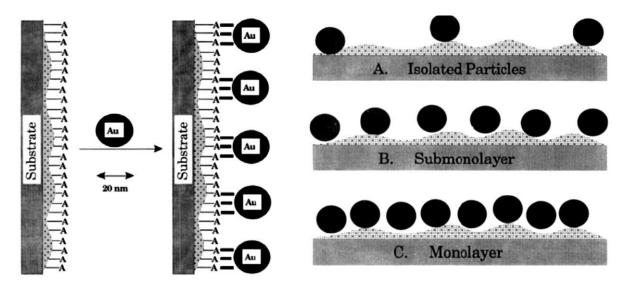


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#### NC arrays fabricated by Chemical self-assembly

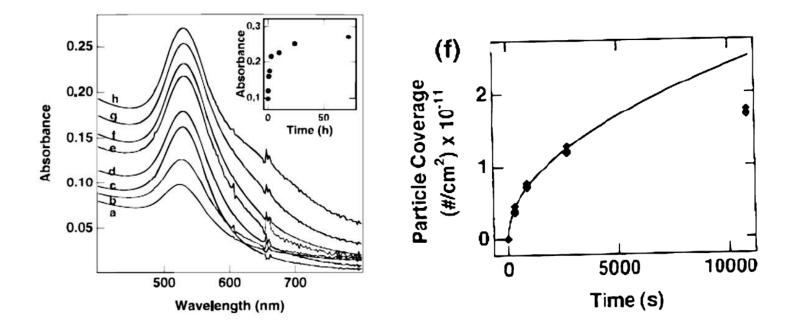


NCs of different sizes have been synthesized from AuCl<sub>4</sub>-, predominantly, using either citrate or sodium borohydride as reducing agents.





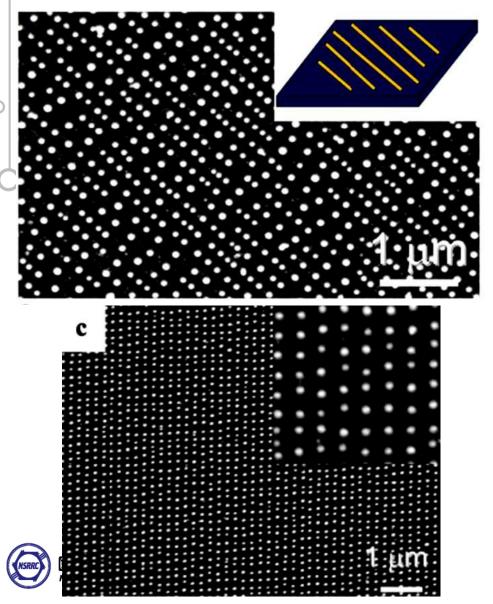
Hydroxyl/oxide groups on the substrate (e.g., glass, quartz, Si) surface provide active sites for the attachment of an alkoxysilane possessing functional group A, where A has a high affinity for gold.

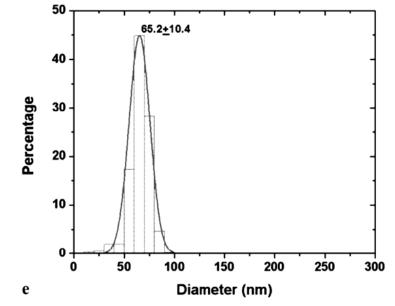


The absorbance spectra and particle coverages for glass slides immersed in 15-nm-diameter colloidal Au solution as a function of time.

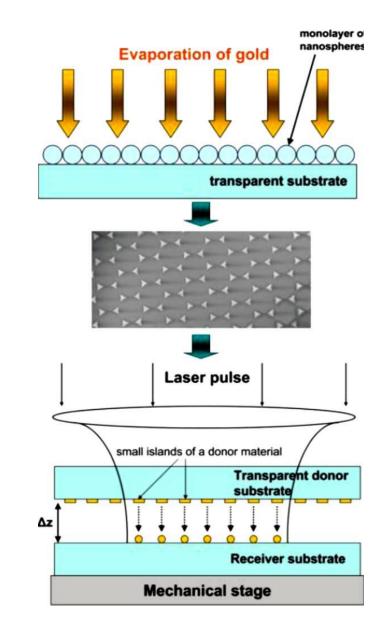


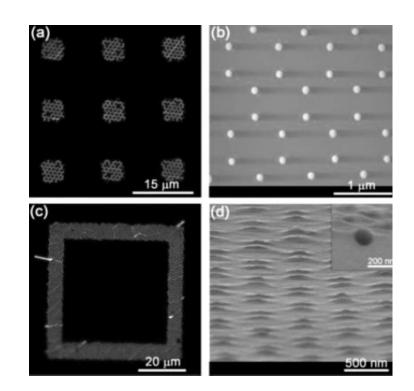
#### Nanocluster arrays fabricated by laser annealing





The method starts with a thin metal lines deposited on a substrate, followed by melting using a single excimer laser pulse.

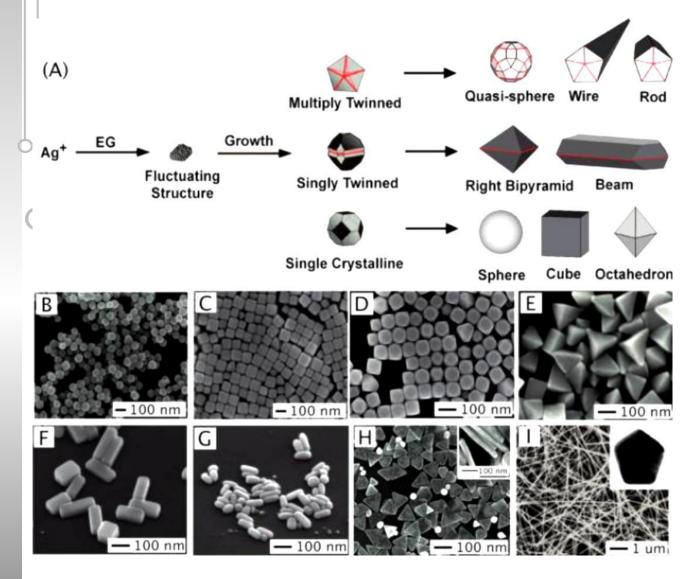




(a) NP structure fabrication by a combination of the nanosphere lithography and laser-induced transfer. (b) Dark-field microscope image and SEM of gold NPs fabricated by single laser pulses on a receiver substrate.

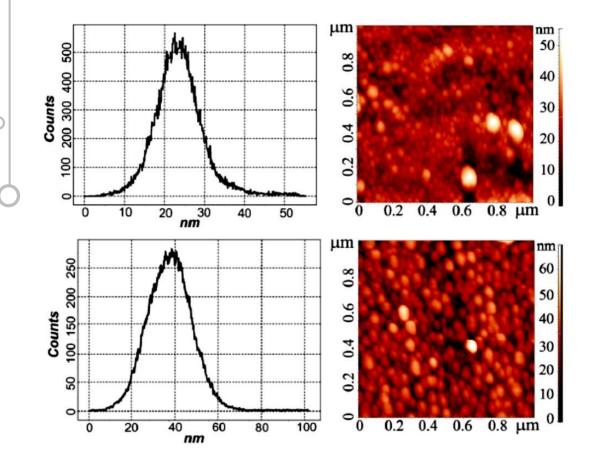


#### Methods of controlling the NPs size



Since NPs exhibit special properties and these properties vary with the size of NPs, controlling the NPs sizes becomes an important issue. Changing the size of NPs in chemical synthesis methods is usually carried out by precisely controlling the synthesis process and concentration of reactants.

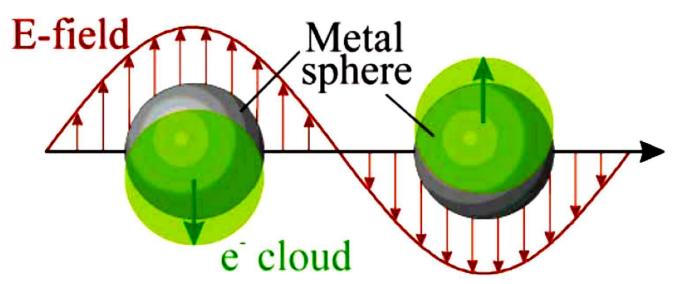




The averaged Ag NP size fabricated under the 45 fs pulse shot was about 20 to 25 nm and those fabricated by 300 ps was about 40 to 50 nm.

國家同步輻射研究中心 National Synchrotron Radiation Research Center Physical and chemical properties of metal NCs

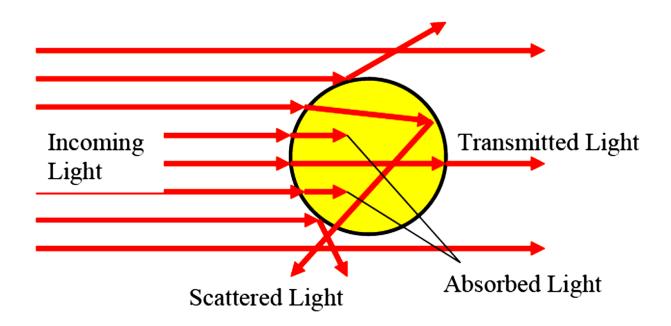
Surface plasmon resonance (SPR)



Schematic diagram of plasmon oscillation for a sphere, showing the displacement of the conduction electron charge cloud relative to the nuclei.



The strong interactions of metallic NCs with incident light, i.e., with the oscillating electric field, originate from the excitation of collective oscillations of conduction electrons within these particles. The collective oscillation of the electrons is called the dipole plasmon resonance of the particle.

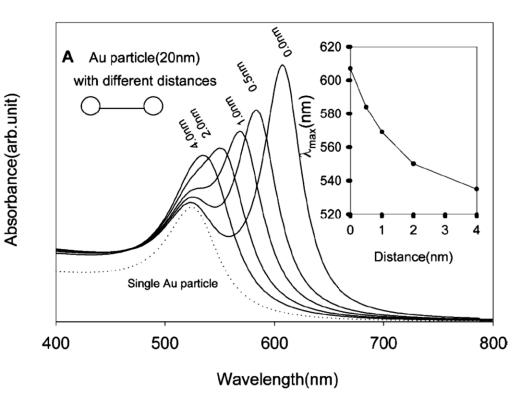


The theoretical principles that describe the metal NCs interaction with light are covered by Mie, Maxwell–Garnett and Drude models.

The extinction spectra (extinction = scattering + absorption) of spherical particles of arbitrary size.



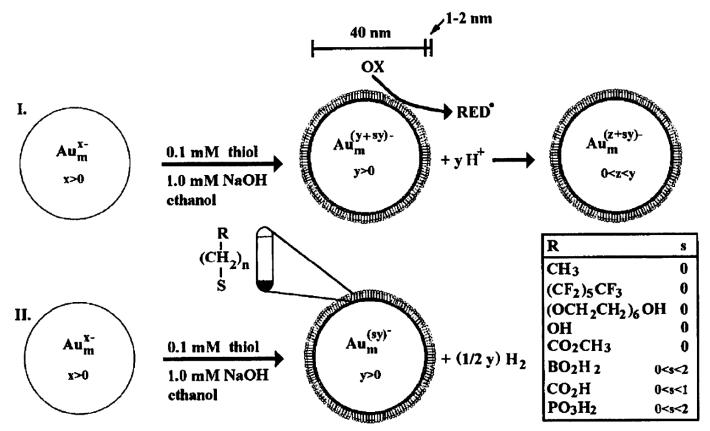
#### Particle-particle coupling



The close contact of metal NCs leads to the appearance of an SPR band attributed to the coupled plasmon absorbances of the NCs. This property has been predicted theoretically.



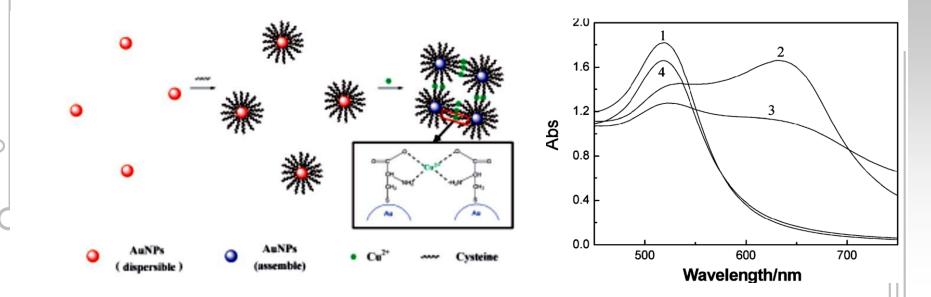
#### Chemical affinity

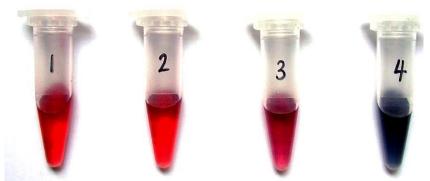


Gold-thiol chemistry: The modulation of SPR for surface-modified Au nanostructures should be dominated by the dielectric constant of the absorbed layers instead of the bulk solvent medium. As a result, Au nanostructures provide a highly sensitive means of detecting changes that occur in the region extremely close to the gold-solution interface.



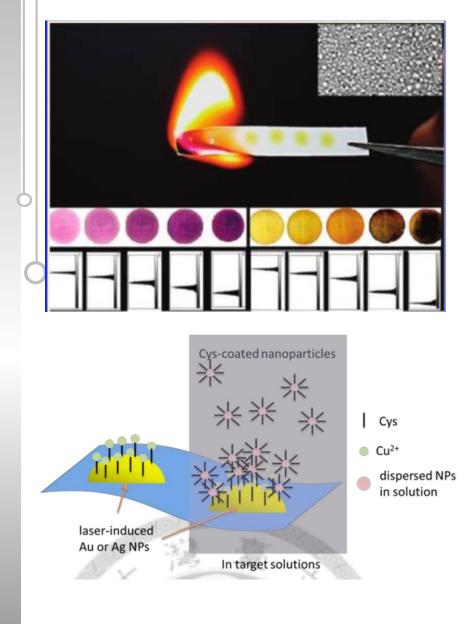
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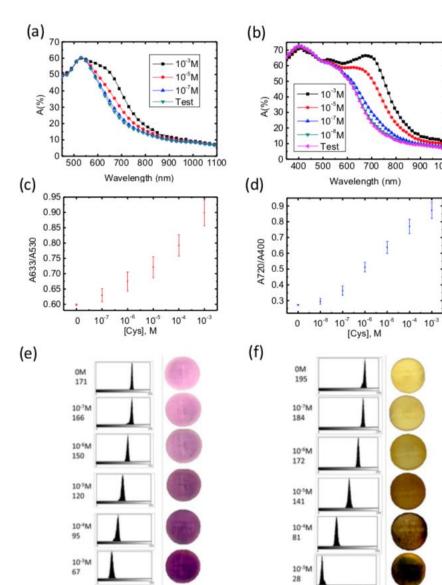




Absorption spectra of AuNPs (1) in the absence of cysteine, (2) in the presence of 1\*10-5M cysteine and 1mM Cu<sub>2+</sub>, (3) in the presence of 1\*10-6M cysteine and 1mM Cu<sub>2+</sub>, and (4) in the presence of 1\*10-7M cysteine and 1mM Cu<sub>2+</sub>.



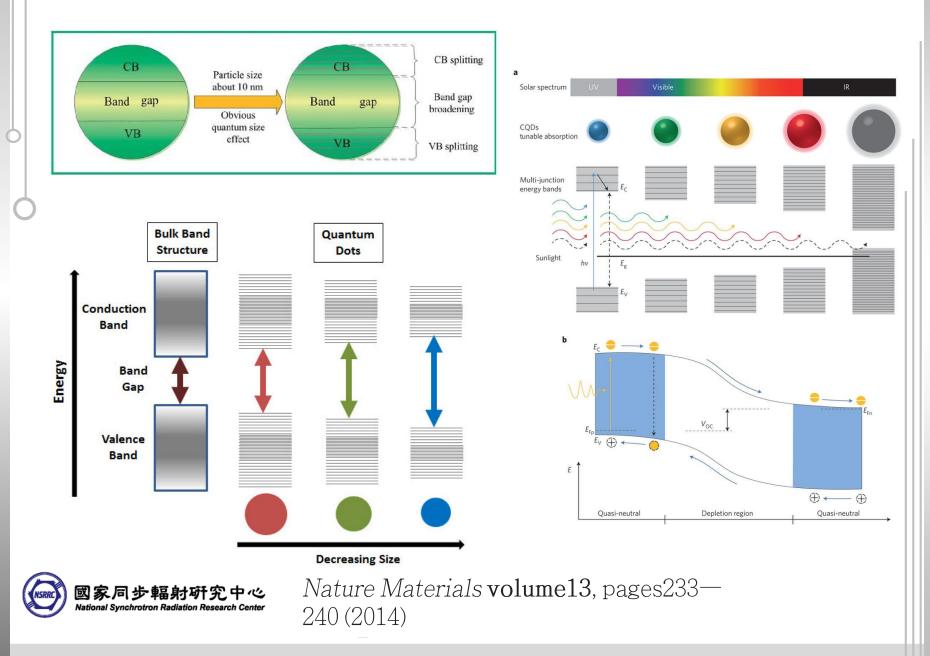




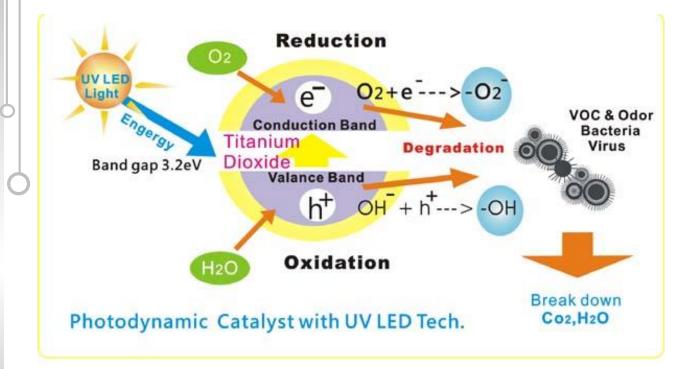
900 1000



#### Quantum size effect



#### Catalytic properties

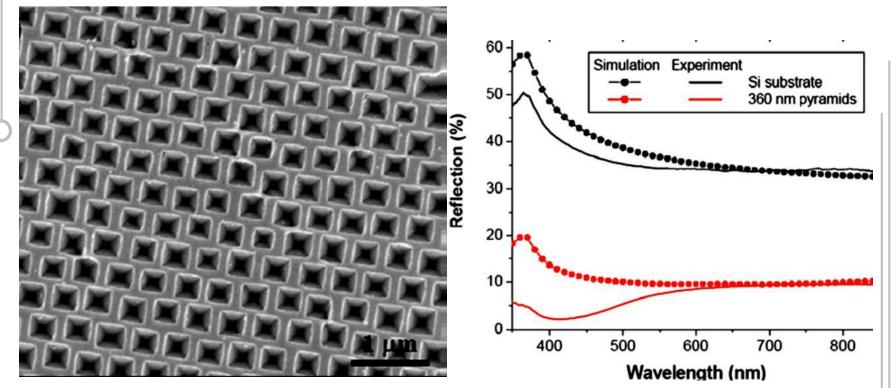


Au NCs supported on Co<sub>3</sub>O<sub>4</sub>, Fe<sub>2</sub>O<sub>3</sub>, or TiO<sub>2</sub> were highly active catalysts, under high dispersion, for CO and H<sub>2</sub> oxidation,<sub>51</sub> NO reduction,<sub>52</sub> water-gas shift reaction,<sub>53</sub> CO<sub>2</sub> hydrogenation,<sub>54</sub> and catalytic combustion of methanol<sub>55</sub> was a surprise, and was considered important by the chemical community.



#### Surface Antireflection

#### Micro-scale texturing techniques



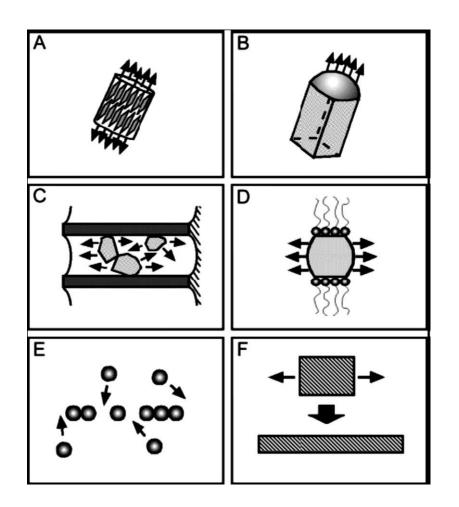
Replicated inverted pyramid arrays in silicon.

Experimental (solid) and simulated (dotted) optical reflectivity at normal incidence.



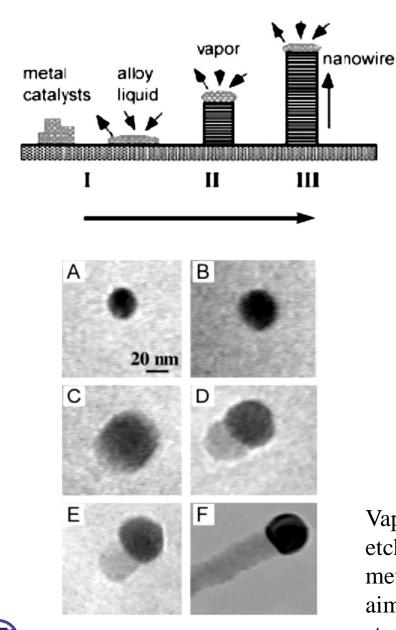
Most of optical devices on thin layer of a dielectric as a antireflective coating to reduce the reflection of light from the front surface of the cell.

#### Sub-wavelength antireflective texturing techniques

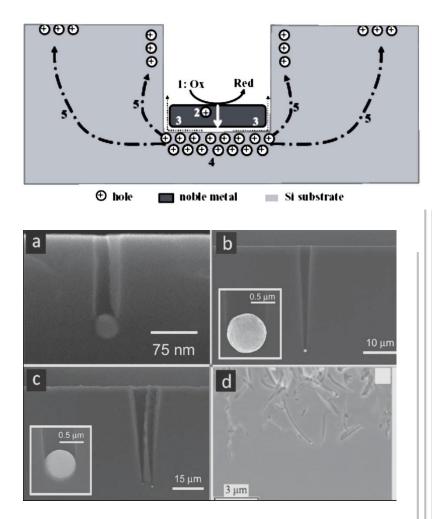


Numerous methods have been developed to fabricate Si nanostructures using top-down or bottom-up approaches



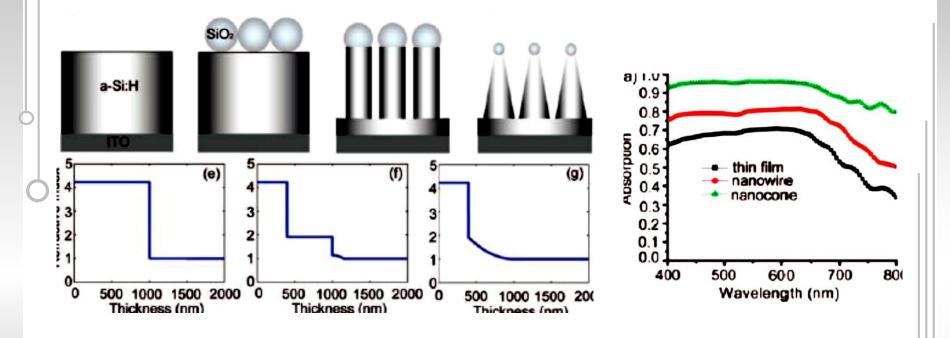


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Vapor-liquid-solid (VLS) growth, reactive ion etching (RIE), electrochemical etching, or metal-assisted chemical etching, all of which aim to control various parameters of the Si structures.

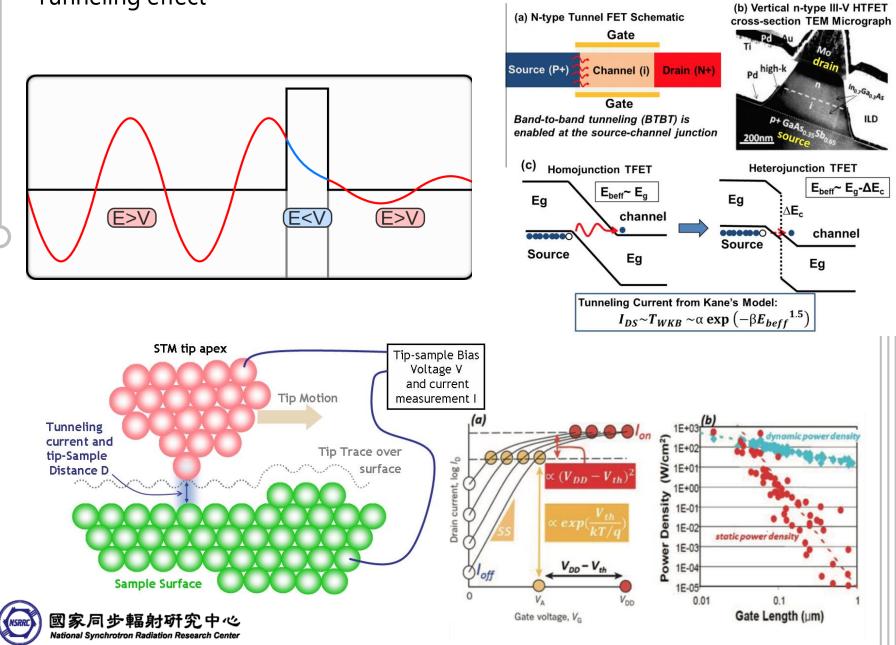
#### One-dimensional nanostructure optical



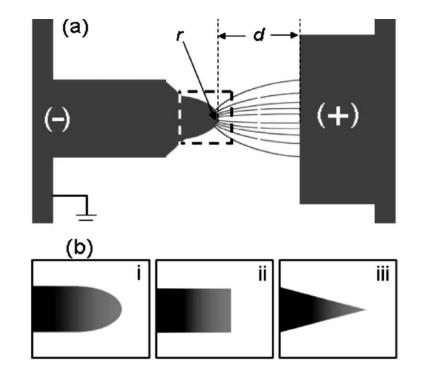
The enhancement effect is particularly strong for Si surface nanostructure arrays, which provide nearly perfect impedance matching between Si and air through a gradual reduction of the effective refractive index.

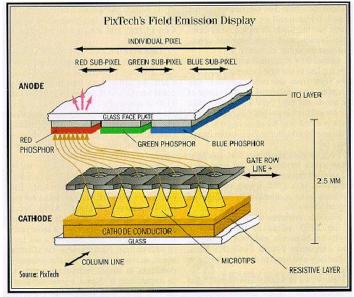


#### Tunneling effect



#### One-dimensional nanostructure field emission properties





PixTech uses a Spindt-type conical cathode structure. The molybdenum cathodes are about 1.2 µm tail. There are hundreds of such cathodes for each pixel, containing red, green, and blue phosphor elements.

Field-emission, is one of the main features of nanomaterials and nanostructures, and is of great commercial interest in displays and other electronic devices.

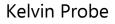


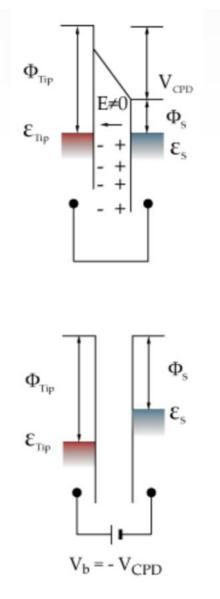
# Outline

- Advantage of materials in nanometer dimensions
- Nanomaterial analysis technology



#### ${\bf An} nalysis \text{-} Work function$

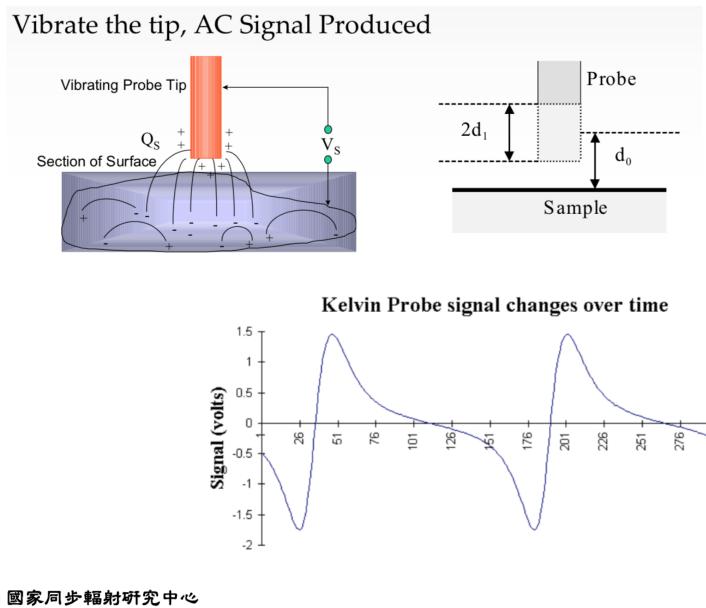




Non-contact, non-destructive vibrating *capacitor* device used to measure the *work function* of conducting materials or *surface potential* of semiconducting or insulating surfaces.

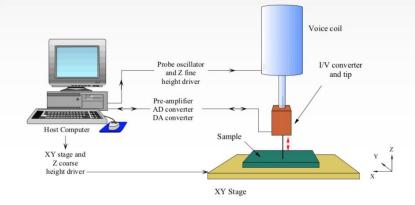
• The technique is extremely sensitive to the topmost layers of atoms or molecules, work function resolution of 1 - 3 meV.

• Unique 'off-null' measurement system also maintains average tip-sample separation to within 1 m, tip to sample tracking



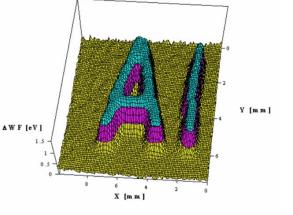
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## Scanning Kelvin Probe 1993



PC houses the digital oscillator (which powers the voice coil actuator), data acquisition system and motorised (x,y,z) stage controller. The signal is derived from a low-noise, high-gain current to voltage (I/V) converter mounted close to the tip.

I.D. Baikie et al Rev. Sci. Instrum. 70, 1842 (1999), Rev. Sci. Instrum 69, 3902 (1998).

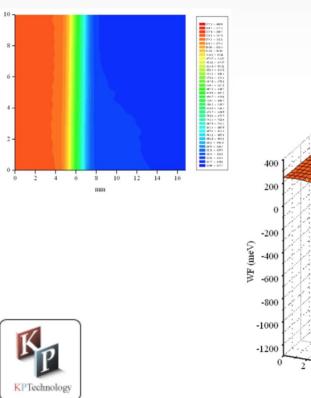






## Gold / Aluminium Reference Sample

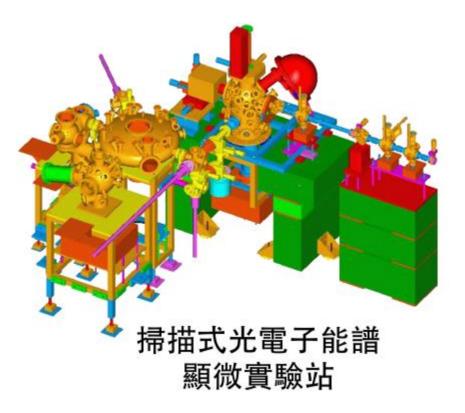
8 10 mm Tip Size: 2mm Scanning Area: 10x16mm Operator: I. Baikie





## 掃描式光電子能譜顯微術

#### (Scanning Photoelectron Microscopy; SPEM)

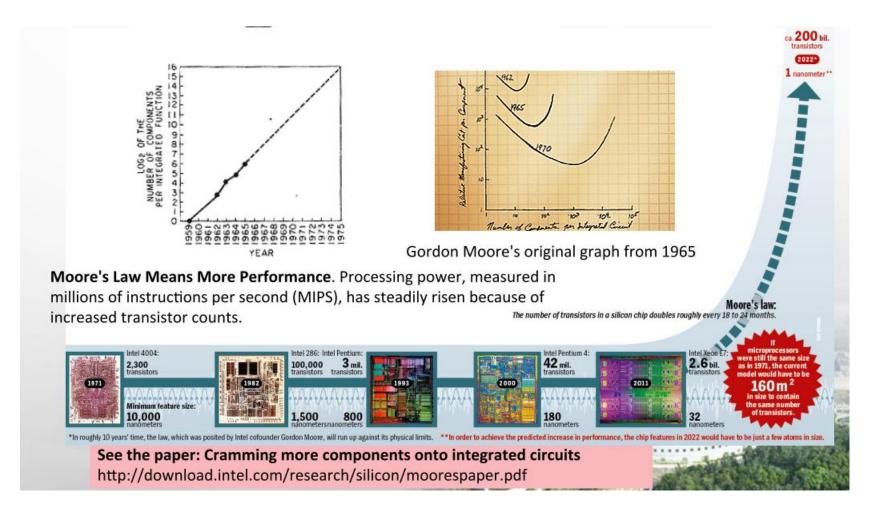






一. 為何發展光電子能譜顯微術?
二. 何謂光電子能譜術?
三. 如何顯微?
四. 舉例說明。
五. 其餘新型態顯微術簡介。







## 為何發展光電子能譜顯微術?

## 小尺度結構分析:STM,TEM,SEM...等

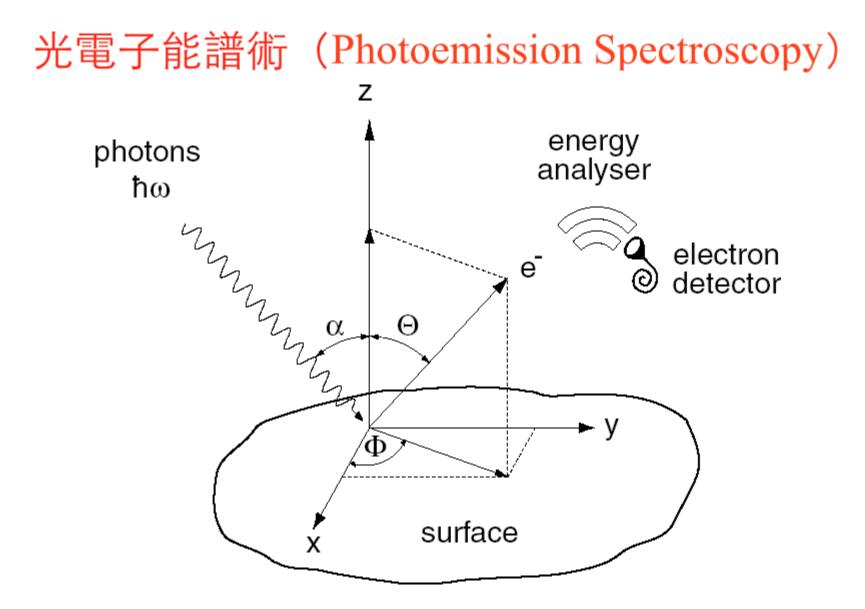
## 小尺度成份分析:





# 二、何謂光電子能譜術?

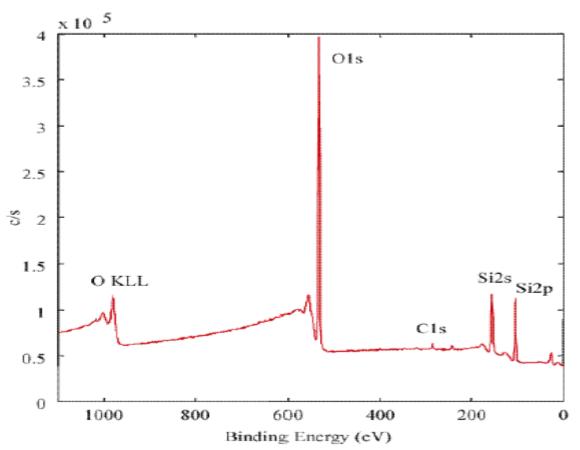


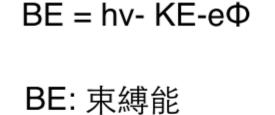




Binding Energy (eV)

Survey Spectrum of Silicon Wafer





hv:入射光能量

KE:電子動能eΦ:功函數

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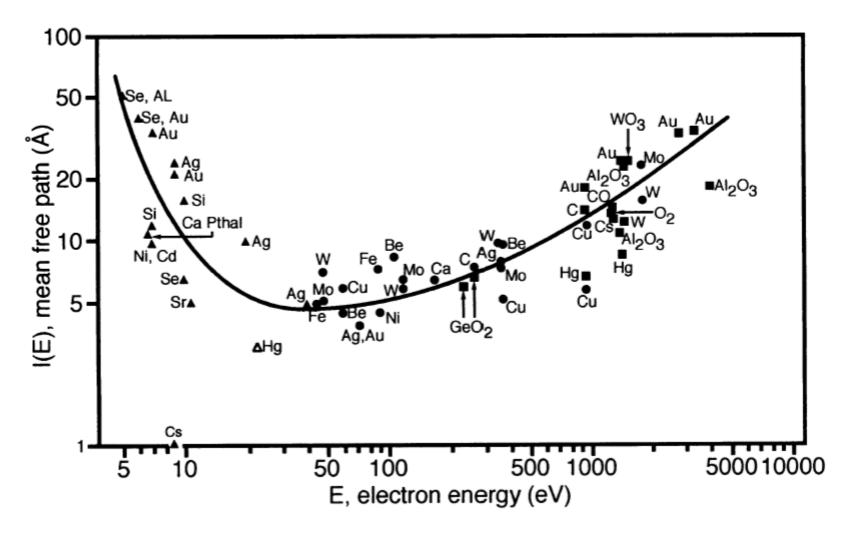
## 典型 ESCA 能譜

## 光電子發射之特性:

# 1.非破壞性 2.表面分析技術 3.具化學鍵結分辨能力



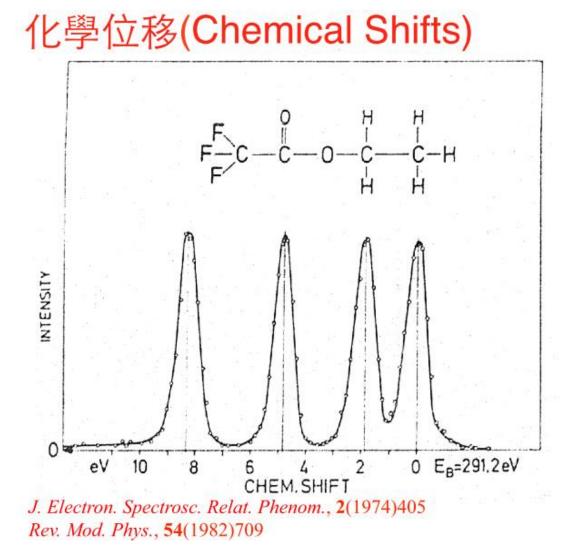
## **Universal Curve**





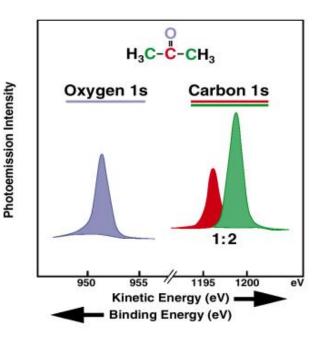
hv e e Source -~µm





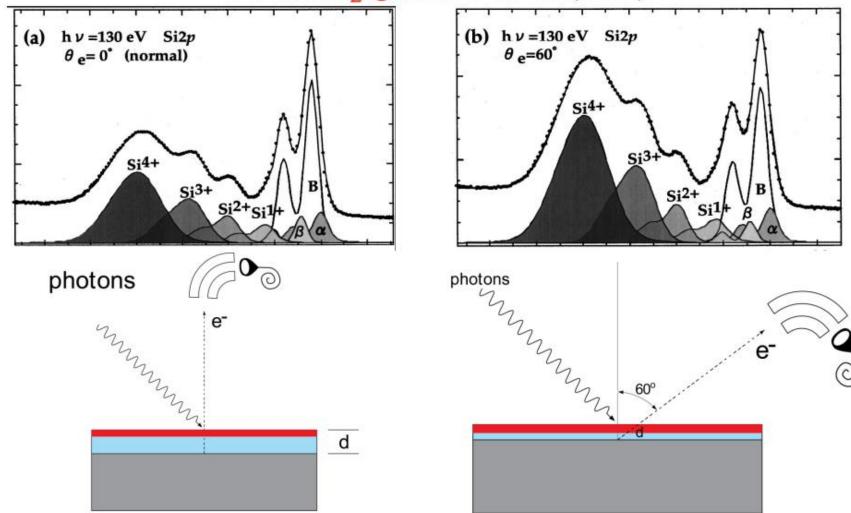
Provides information about

- Kind of atom
- Number of atoms
- Chemical shift



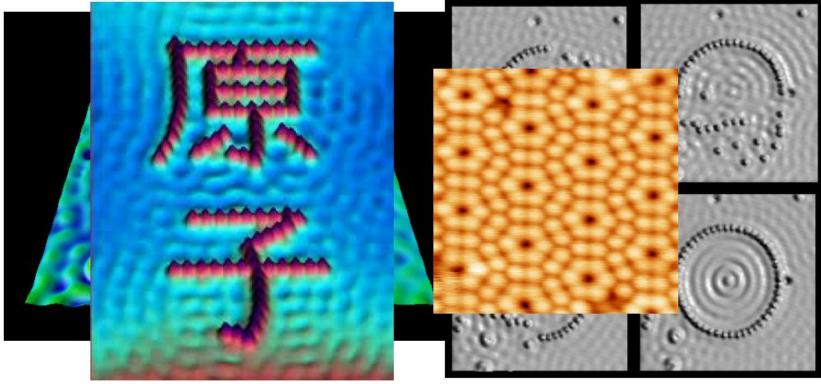


## $6 \text{ Å SiO}_2 \text{ grown on Si}(100)$





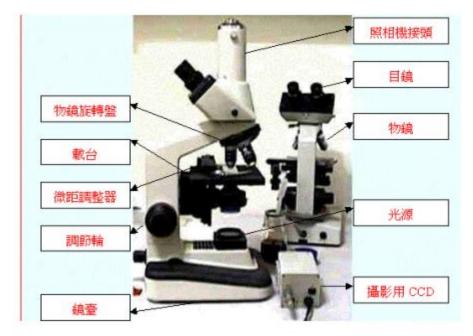
### Scanning Tunneling Microscope (STM) Scanning Probe Microscope (SPM)

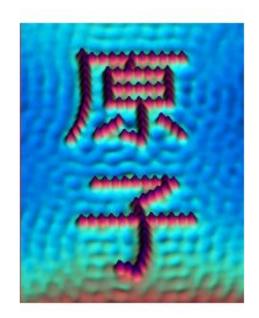




# Almost every analytical technique can benefit from spatial resolution!

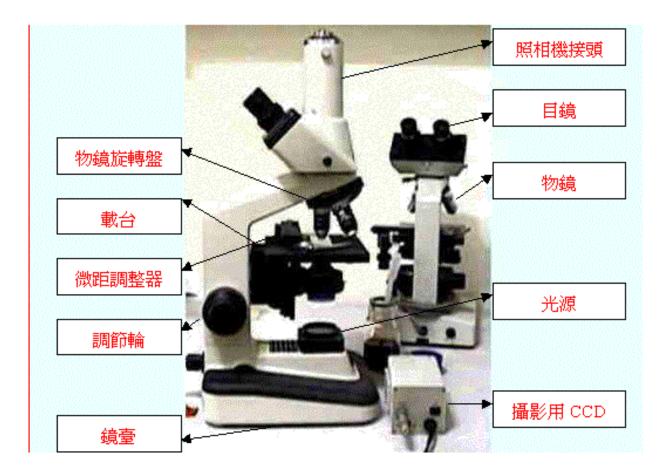
But how to achieve that?





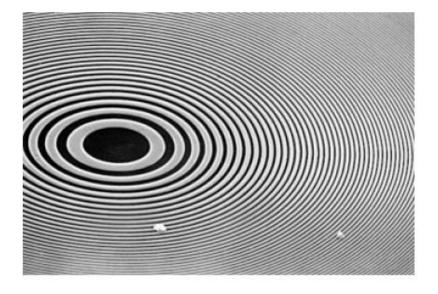


## Does this work?





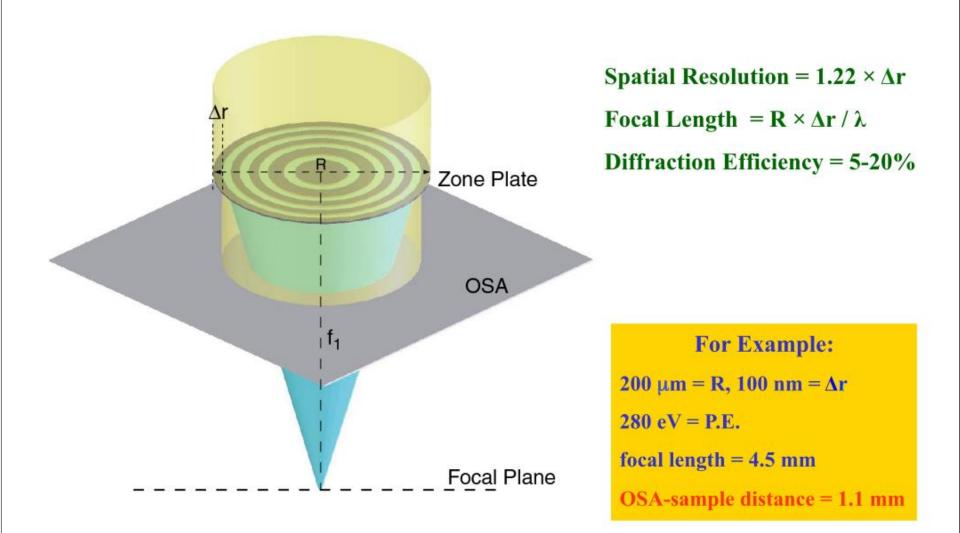
## Fresnel Zone Plate (波帶環片)



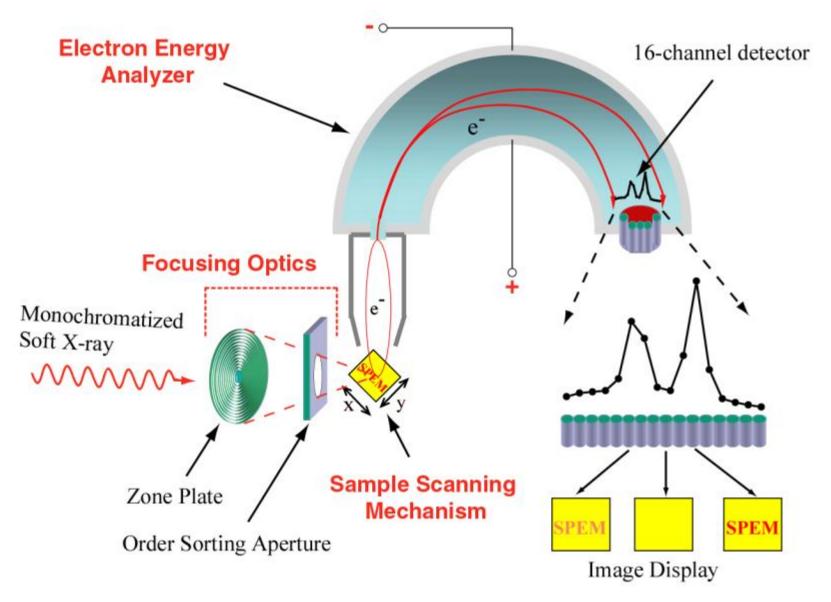


nn NO Si<sub>3</sub>N<sub>4</sub> Si

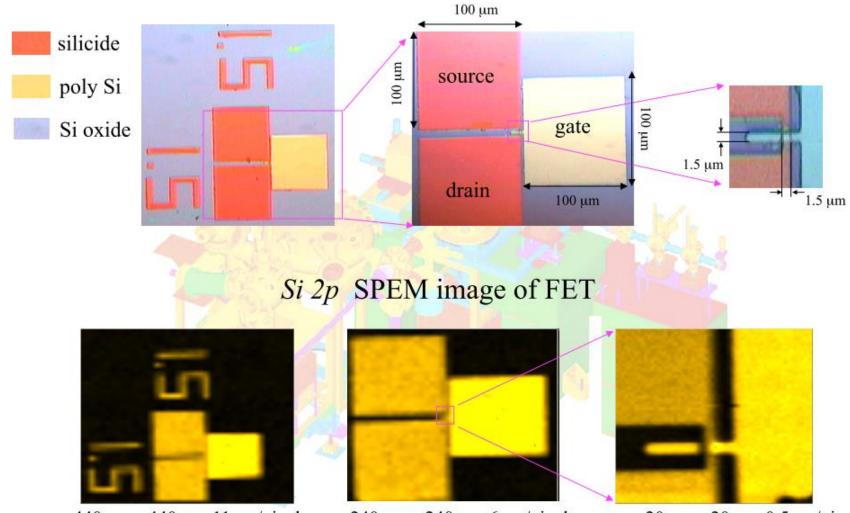












 $440\mu m \times 440\mu m$ ,  $11\mu m/pixel$ 

240μm × 240μm, 6μm/pixel

 $20\mu m \times 20\mu m$ ,  $0.5\mu m/pixel$ 



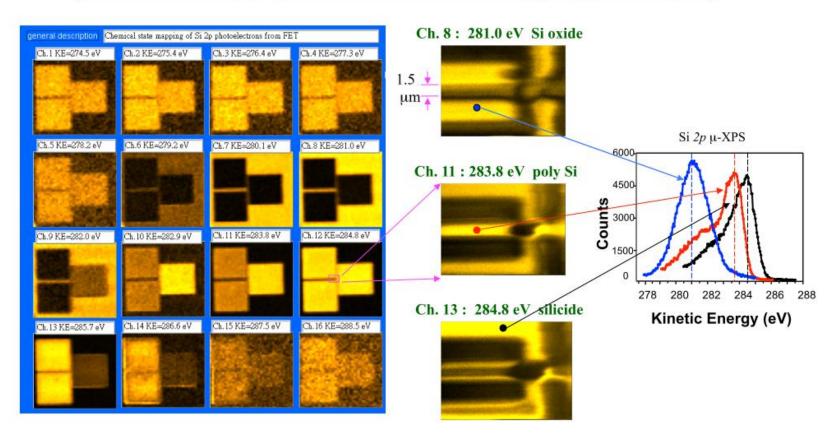
Yes, good image is beautiful! BUT, most of the time,

IMAGES SUGGEST,

SPECTRA TELL



#### Parallel Imaging for Chemical State Mapping (PICSM)





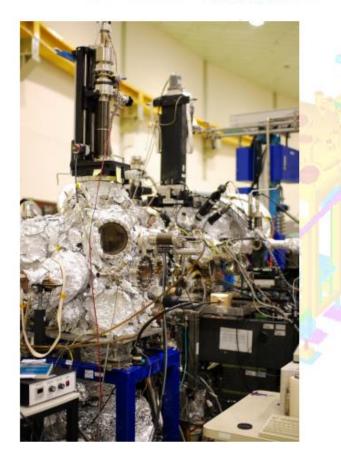
 $12 \ \mu\text{m} \times 12 \ \mu\text{m}, 0.1 \ \mu\text{m/pixel}$ 

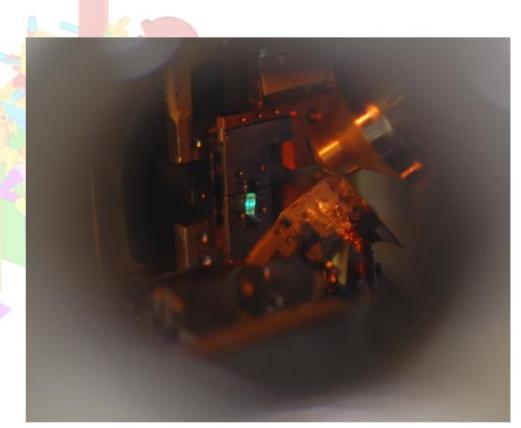


\* \*

\* \*









# **J**、舉例說明

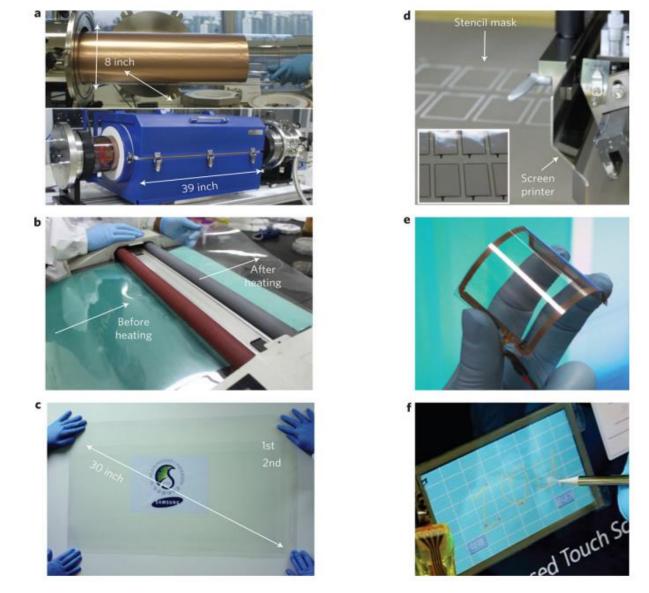


# Is silicon dioxide essential to make graphene visible?









S. Bae et al., Nat. Nanotech., 5, 574 (2010)



Nanotechnology 21 (2010) 175201 (6pp)

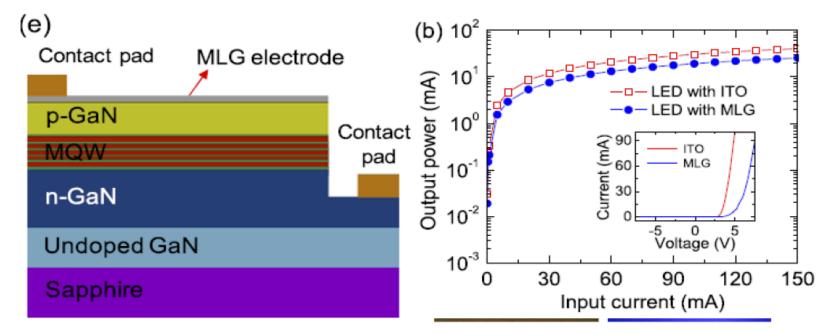
### Large-scale patterned multi-layer graphene films as transparent conducting electrodes for GaN light-emitting diodes

Gunho Jo<sup>1</sup>, Minhyeok Choe<sup>1</sup>, Chu-Young Cho<sup>1</sup>, Jin Ho Kim<sup>3</sup>, <sup>1</sup> Department of Materials Science and Technology, Gwangju 500-712, Korea <sup>2</sup> Department of Nanobio Materials and Technology, Gwangju 500-712, Korea <sup>3</sup> Department of Nanobio Materials and Technology, Gwangju 500-712, Korea <sup>3</sup> Department of Chemistry and SKKU Swarbwar University.

<sup>1</sup> Department of Materials Science and Engineering, Gwangju Institute of Science and Technology, Gwangju 500-712, Korea

<sup>2</sup> Department of Nanobio Materials and Electronics, Gwangju Institute of Science and Technology, Gwangju 500-712, Korea

<sup>3</sup> Department of Chemistry and SKKU Advanced Institute of Nanotechnology, Sungkyunkwan University, Suwon 440-746, Korea





## SLG, BLG or MLG ?

## Multi domain, defect boundaries ?

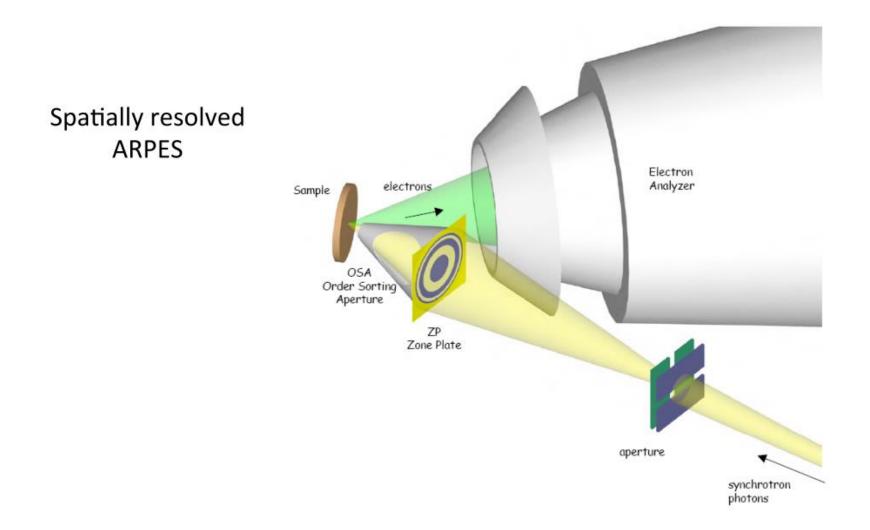


## **Mechanical Exfoliation**



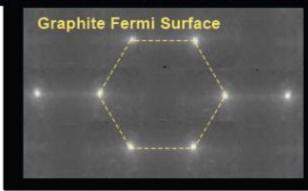
## 新型態同步輻射顯微術簡介

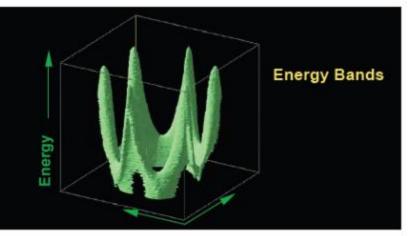






#### conventional ARPES on a large, pure single crystal





most of the momentum information is lost

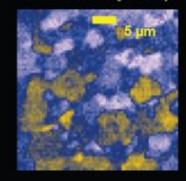
as our spot size is much larger than the grain size.

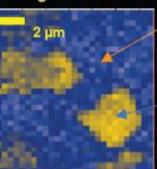
#### conventional ARPES of polycrystalline graphite

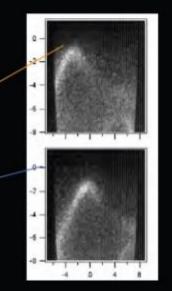
Fermi Surface

#### nanoARPES of polycrystalline graphite

we can recover all the momentum information by sampling one grain at a time

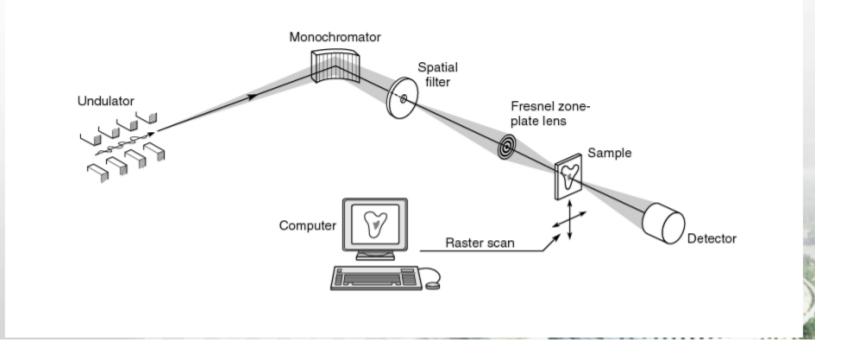








## 掃描式穿透光吸收能譜顯微術 (Scanning Transmission X-Ray Microscope; STXM)





## STXM specifications:

spatial resolution: 25~50 nm

- energy resolution< 0.1 eV</li>
  - sample : 01.-1 micron

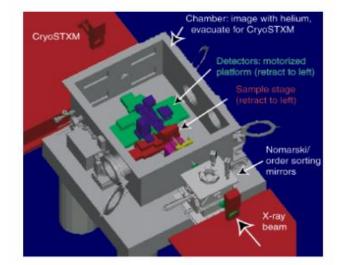
## He/inert gas atmosphere or vacuum environment

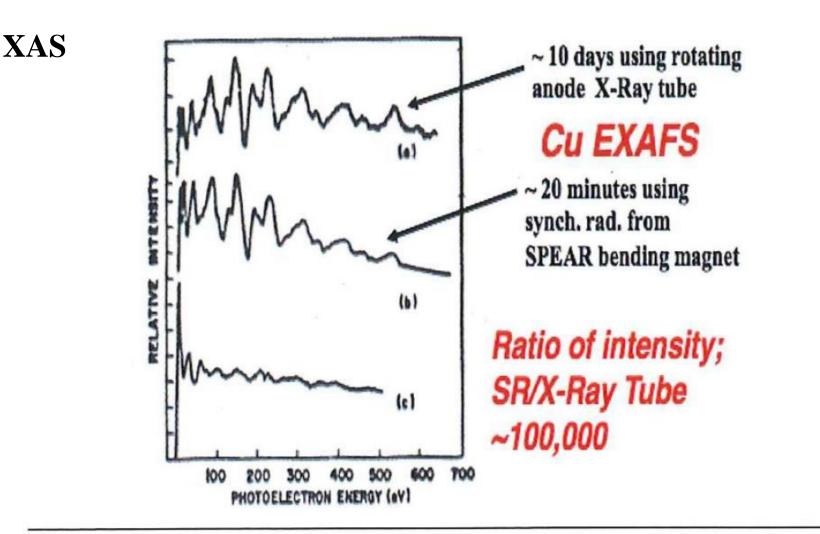
solid/liquid/gas sample

### **Applications:**

- Magnetism
- Soft Matter
- Earth/Environmental Science
- Polymer Science
- Materials Science
- Catalyst







Extended X-ray Absorption Fine Structure for Cu (a) and (b), by Peter Eisenberger and Brian Kincaid, taken in 1974 at the Stanford Synchrotron Radiation Project. (c) is EXAFS from a thin Nb3Ge superconducting thin film.

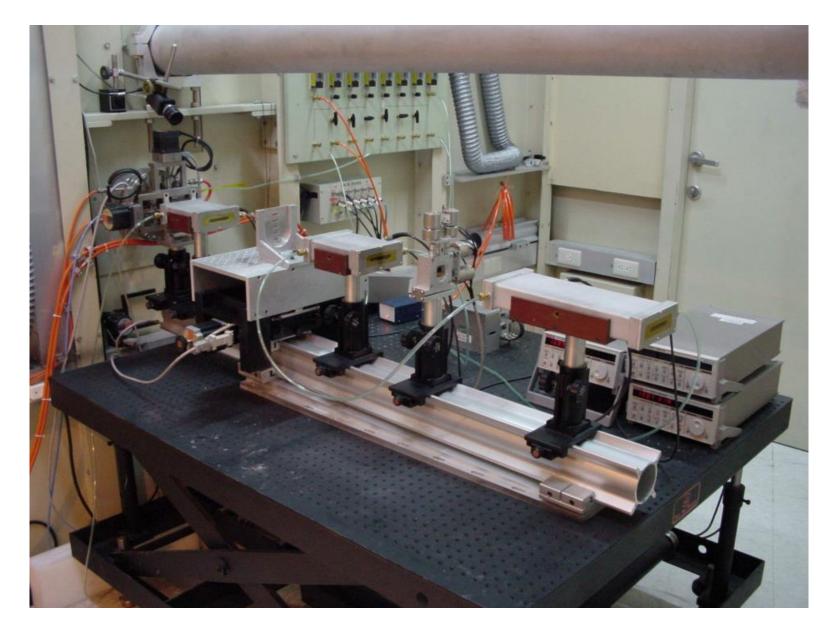


### 典型的 X 光吸收光譜實驗配置圖 (虛線表示實驗站的輻射屏蔽屋)

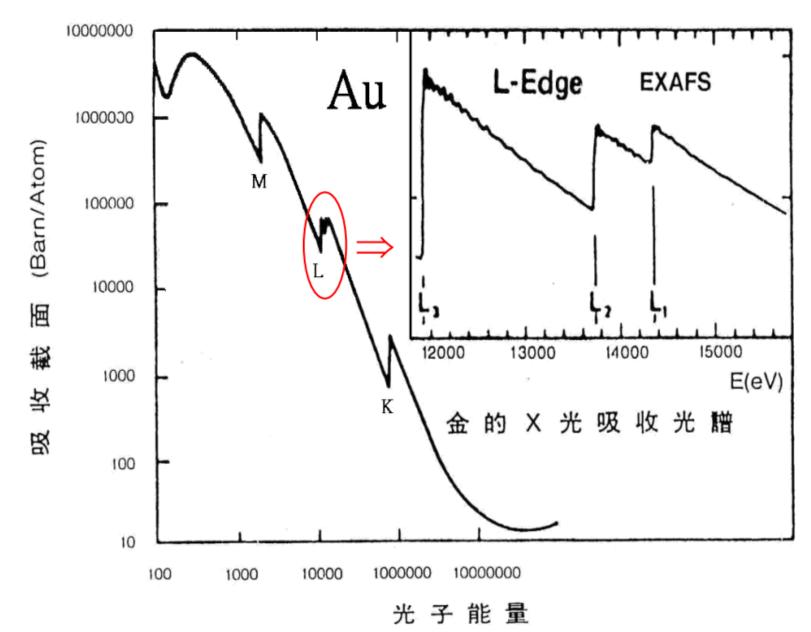
IONIZATION CHAMBERS REFERENCE SAMPLE Ir It FILTER INTERVIEWED IONICHROMATOR If SCATTER SLITS ENTRANCE SLITS

> 穿透法之樣品吸收係數為  $Ln(I_0/I_t)$ 螢光法或電子產率法為  $I_f/I_0$  或  $I_e/I_0$ 參考標準品之吸收係數為  $Ln(I_t/I_r)$

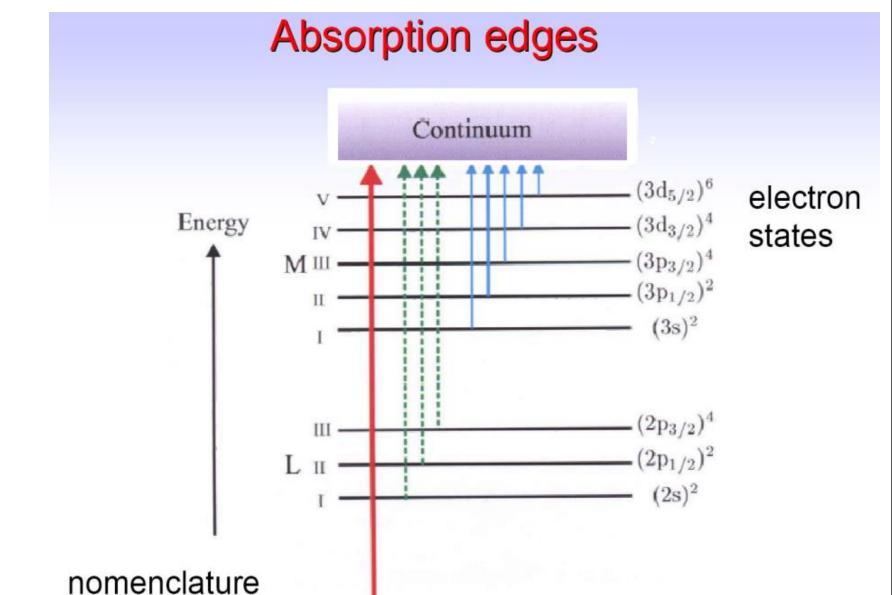












of edges

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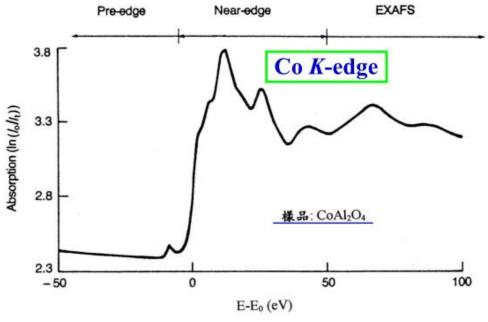
66

 $(1s)^2$ 

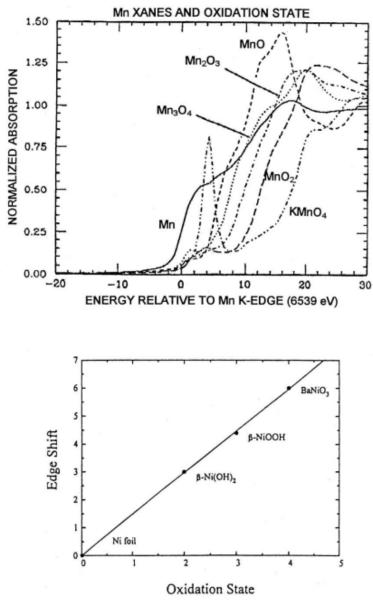
X 光吸收近邊緣結構 (XANES) ====> 吸收原子的電子性質(如:氧化價數以及 d-軌域之電子填滿程度) 吸收原子所處之晶位對稱性

延伸 X 光吸收精細結構 (EXAFS) ====> 吸收原子周圍之局部幾何結構(如:周圍原子種類、配位數、 原子間距離、排列之雜亂度)

> EXAFS: Extended X-ray Absorption Fine Structure XANES: X-ray Absorption Near-Edge Structure





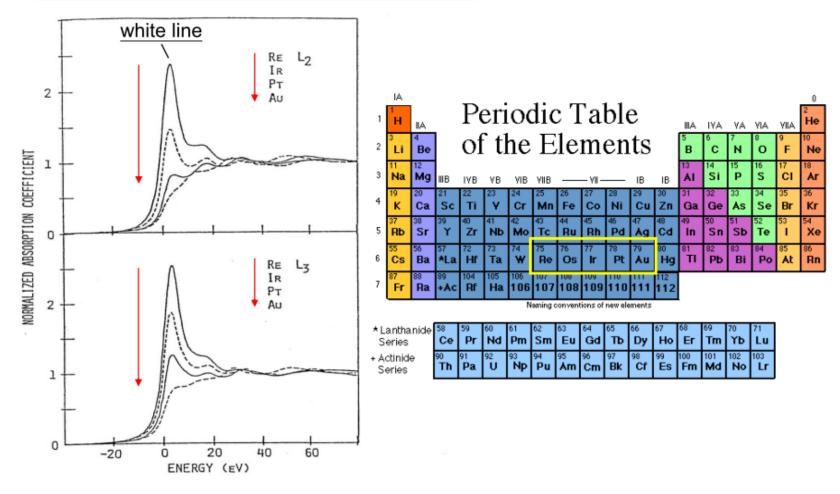


- 1. Absorption edges of many elements show significant energy shifts with <u>oxidation state</u>.
- 2. Why does edge shift with oxidation state?
- Electrostatics harder to remove bound electron.
- Higher oxidation states have shorter bonds (in general).

Fig. 3. Plot of the edge shifts in the XANES spectra vs. the increasing oxidation state of nickel.

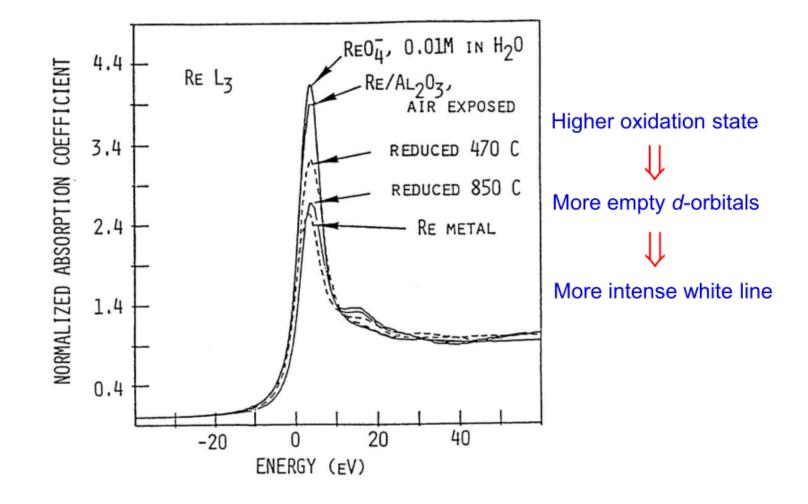


### "White line" intensity reflects the d-orbital occupancy



XANES spectra for the  $L_{2,3}$ -edges of Re, Ir, Pt, and Au. Spectra were normalized to unit edge jump and aligned to the first inflection point.





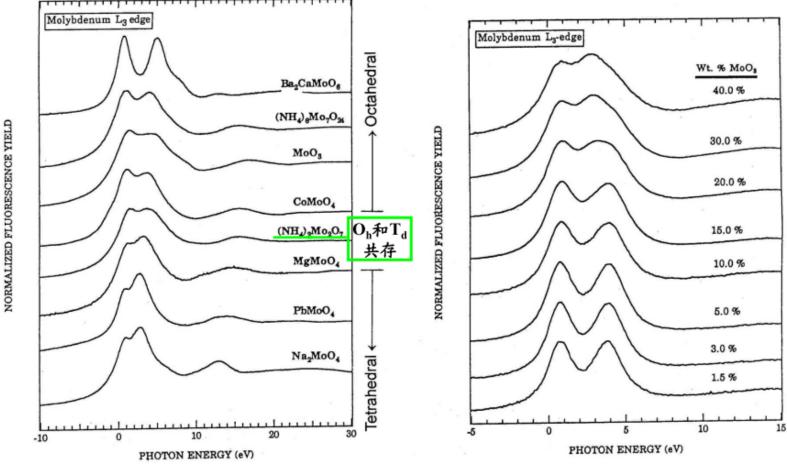
Overplot of Re  $L_3$ -edges for 1 wt% Re on Al<sub>2</sub>O<sub>3</sub>

catalyst with Re<sup>+7</sup> and Re metal reference edges.



### Why are we interested in XANES?

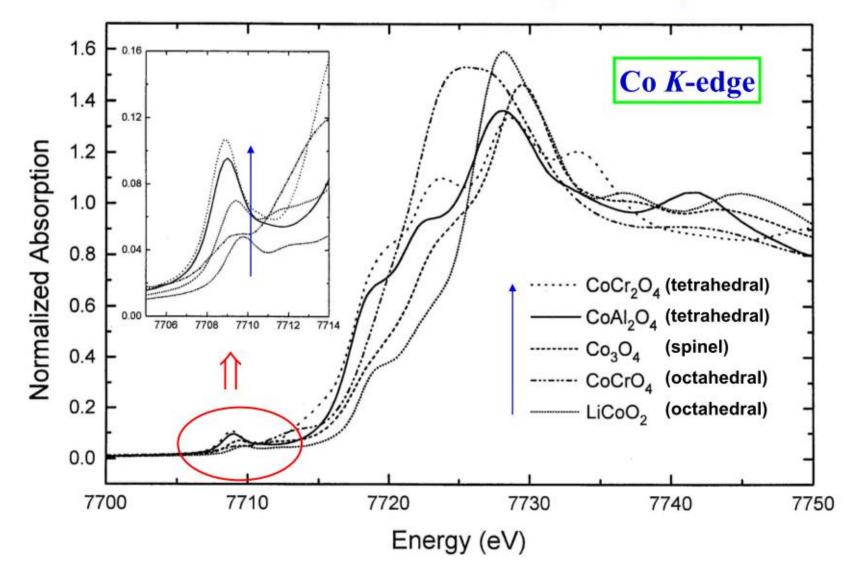
### Local Coordination Symmetry





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### **Local Coordination Symmetry**





## **Theory of EXAFS**

當中心原子A的內層電子因吸收X光而被游離時,此種光電子(photo-electron)將帶著E-E<sub>0</sub>的動能遠離原子核,形成一向外行進的光電子波,其波長為: h:蒲朗克常數

$$\lambda_e = \frac{h}{p} = \frac{h}{\left[2m(E-E_0)\right]^{\frac{1}{2}}}$$
p:電子動量
m:電子質量

若吸收原子周圍存在其他原子B時,會將向外行進的光電 子波予以**背向散射**,假設A、B兩原子相距R,則向外行進 與背向散射的光電子波之間存在2R的路程差,此一路程差 將使得二者的相位差為:

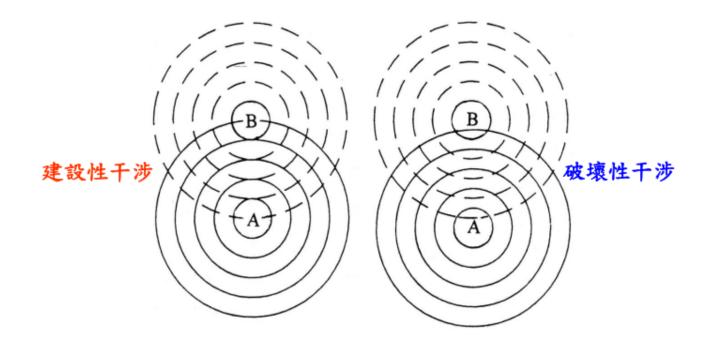
$$2\pi \left(\frac{2R}{\lambda_e}\right) = 2R \left[\frac{8\pi^2 m(E-E_0)}{h^2}\right]^{\frac{1}{2}} = 2kR$$

其中k稱為**光電子波向量**,常以 $A^{-1}$ 為單位。 當電子動能的單位為eV時, $k = [0.2625(E - E_0)]^{\frac{1}{2}}$ 

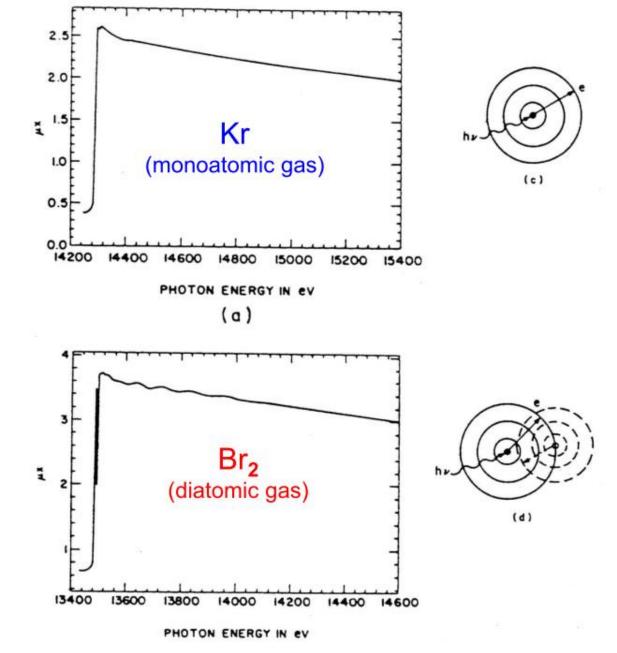


## Theory of EXAFS

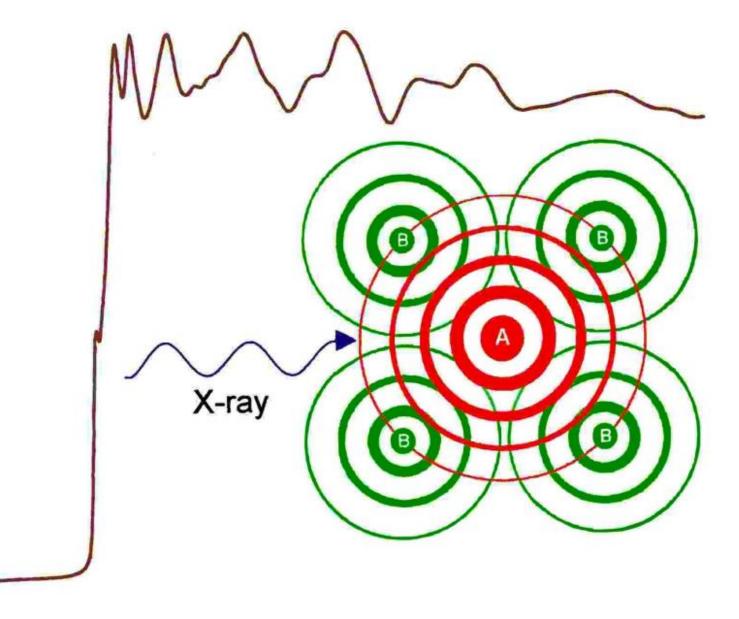
向外行進與背向散射的光電子波彼此間的相位差將隨原子間距離 及入射能量而變化,進而產生建設性(同相)或破壞性(反相)干涉, 造成在吸收係數上之調諧作用(modulation),亦即EXAFS振盪。



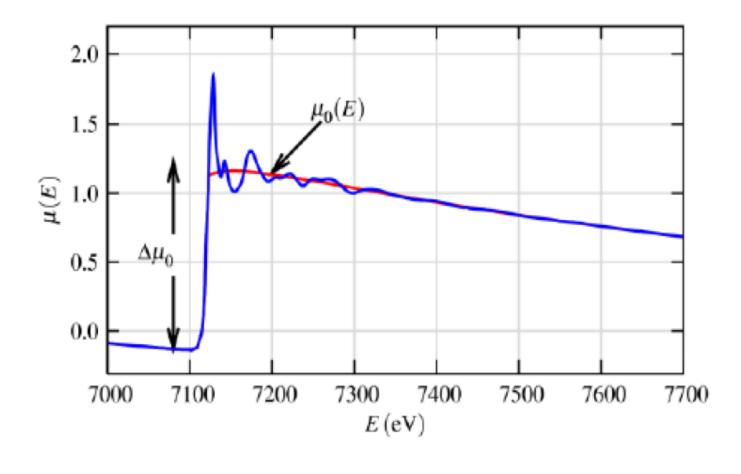








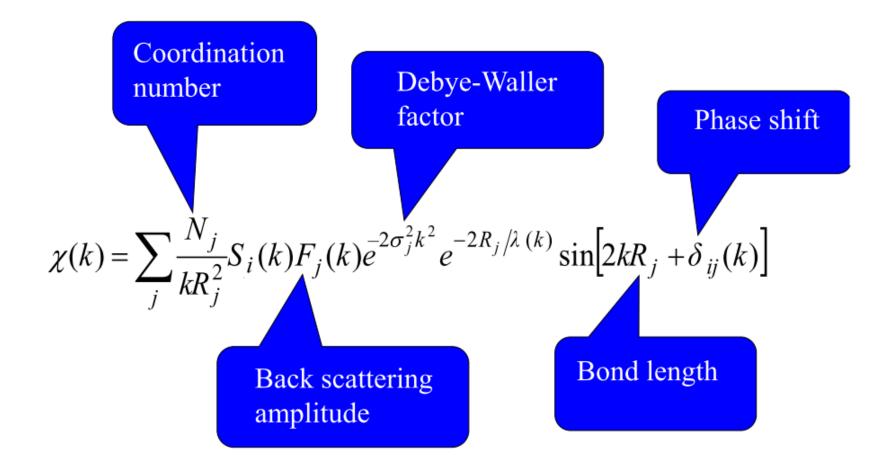




$$\chi(\mathbf{E}) = \frac{\mu(\mathbf{E}) - \mu_0(\mathbf{E})}{\Delta \mu_0(\mathbf{E}_0)} \qquad \qquad \mathbf{k} = \sqrt{\frac{2\mathbf{m}(\mathbf{E} - \mathbf{E}_0)}{\hbar^2}}$$



# EXAFS函數與各結構參數間的關係式



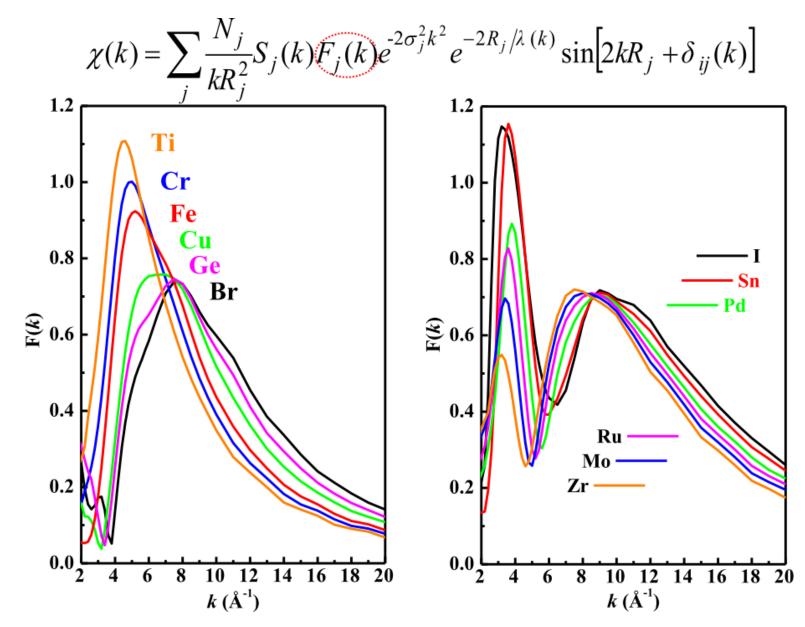


# Structural Parameters from EXAFS Analysis

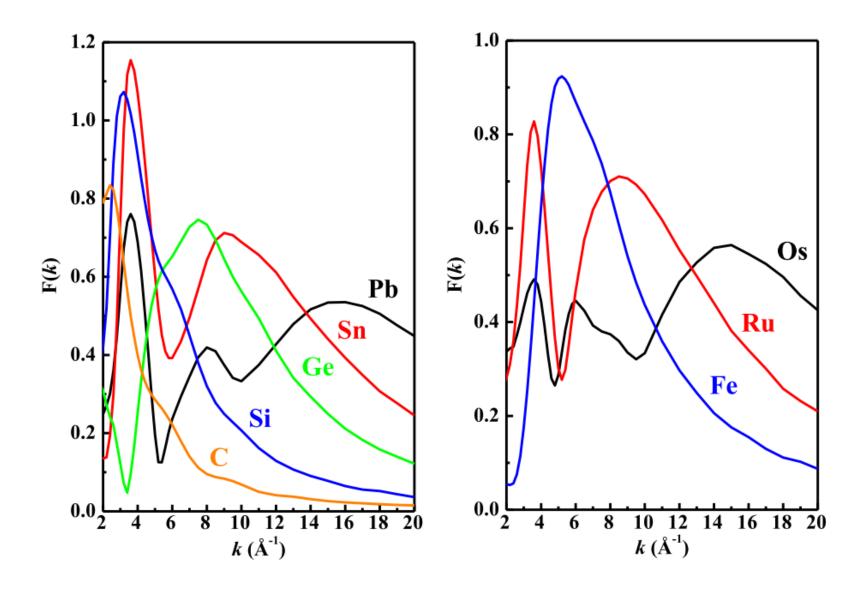
Structural parameter	Accuracy	Observable spectral features
bond length (interatomic distance)	±1%	frequency
coordination number	$\pm 20\%$	magnitude
type of coordination atoms	$\pm 4$ (in atomic no.)	amplitude envelope and phase shift
Debye-Waller factor $(\sigma^2 = \sigma_s^2 + \sigma_t^2)$	±20% (?)	oscillation damping speed

對於每一個配位層皆可透過 EXAFS 數據分析 獲得上表中的四種結構參數,以及ΔE<sub>0</sub>值。





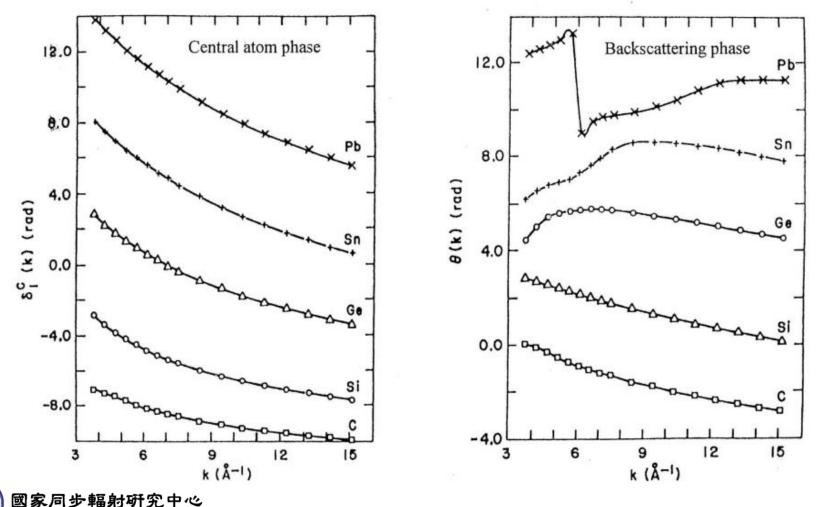






$$\chi(k) = \sum_{j} \frac{N_{j}}{kR_{j}^{2}} S_{j}(k) F_{j}(k) e^{-2\sigma_{j}^{2}k^{2}} e^{-2R_{j}/\lambda(k)} \sin\left[2kR_{j} + \delta_{ij}(k)\right]$$

Fotal phase shift experienced by the photoelectron is given by  $\delta_{ij}(k) = 2 \delta_i^c(k) + \theta(k)$ .



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#### (二) EXAFS 區域內平滑背景的扣除

(三) 正規化處理

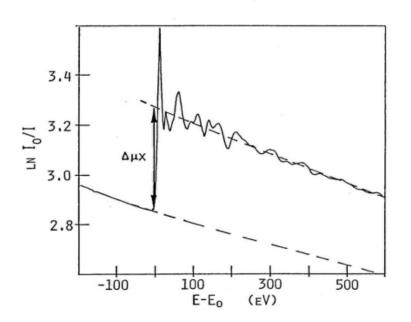


Fig. 6 Illustration of procedure for subtracting the pre-edge and post-edge backgrounds and dividing by the edge jump. The spectrum is measured at Ni K-edge using NiO as the sample.



#### (四)將能量座標轉換至 k-空間並進行 k 之加權運算

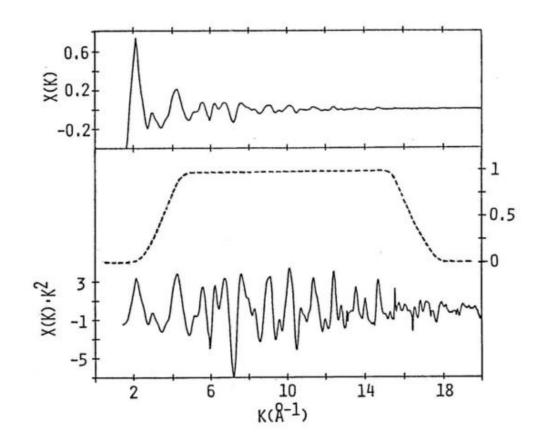


Fig. 7 Background-subtracted and normalized EXAFS function from Fig.6, multiplied by  $k^n$  (n = 0 and 2). Dotted curve represents the 40 % Hanning window function.



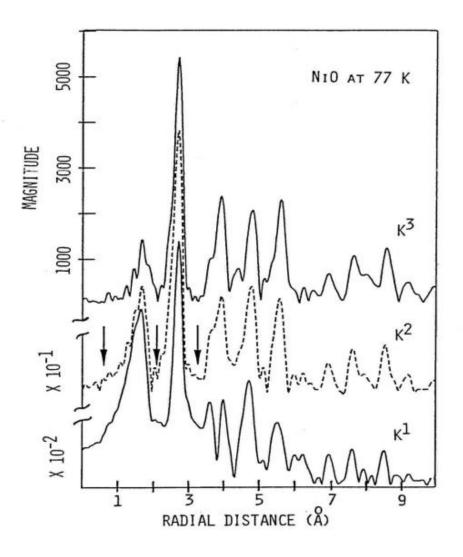
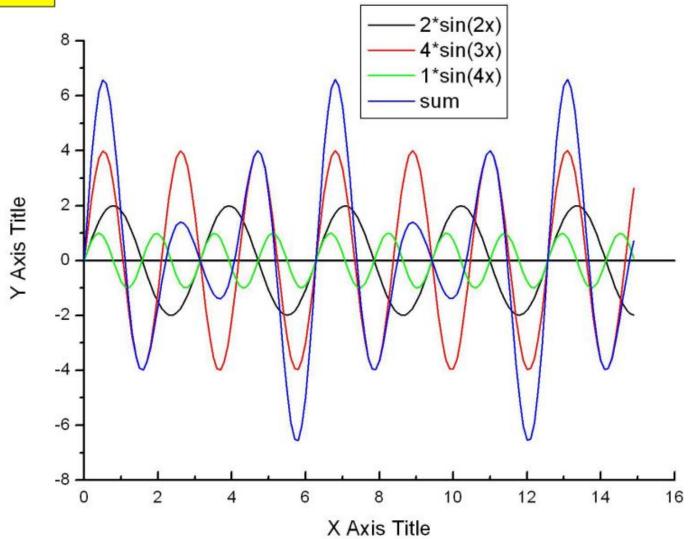


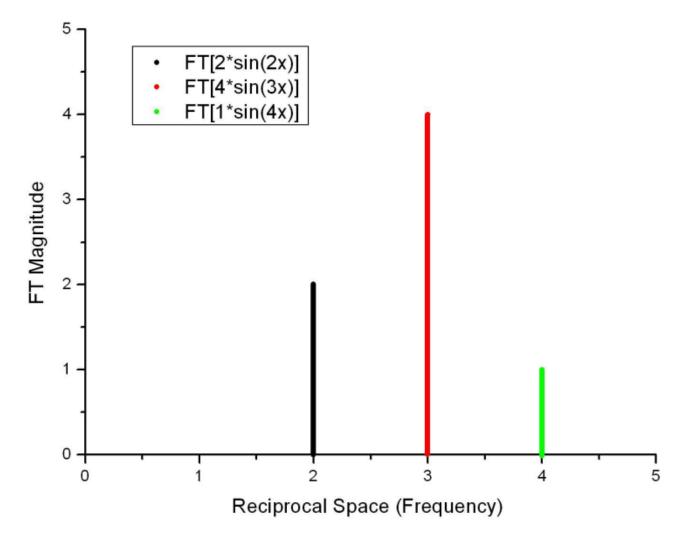
Fig. 8 Fourier transform of  $k^n$ -weighted NiO EXAFS. Vertical arrows indicate limits for inverse transform.













#### (六) 反傅立葉變換

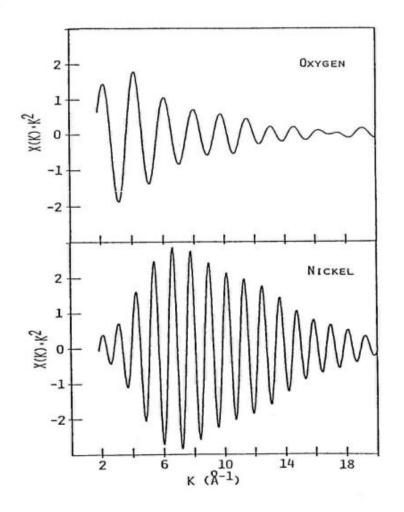
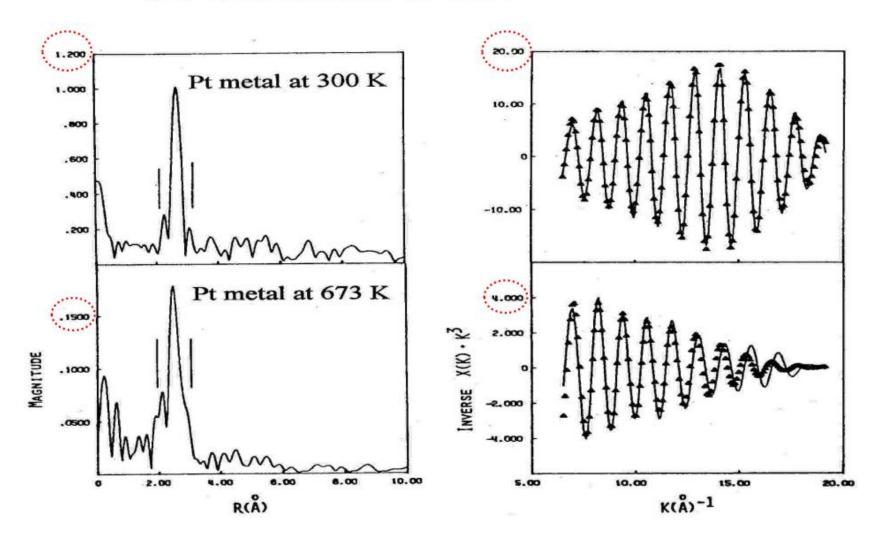


Fig. 9 Inverse transforms for the first two shells in NiO.

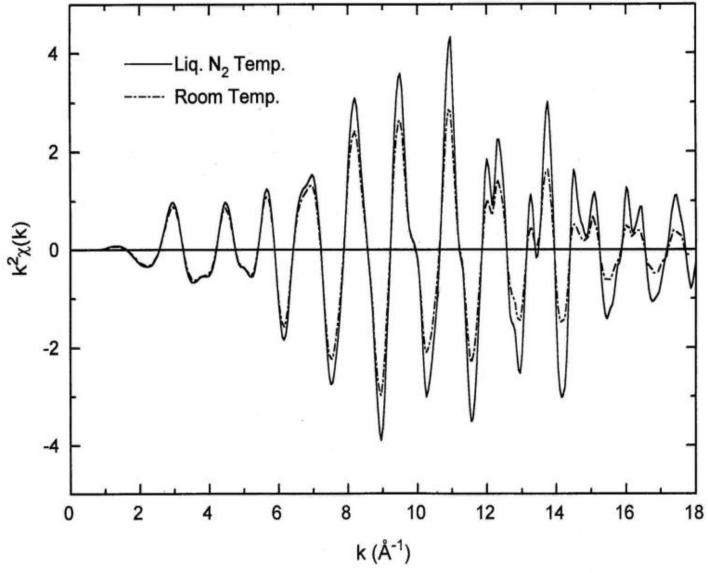


(七) 各配位層結構參數的計算



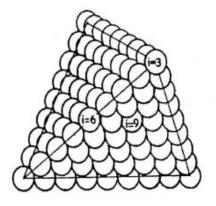




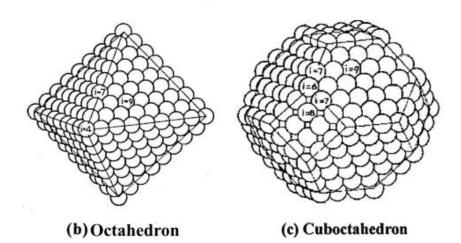




具有fcc晶格結構之三種不同顆粒形狀

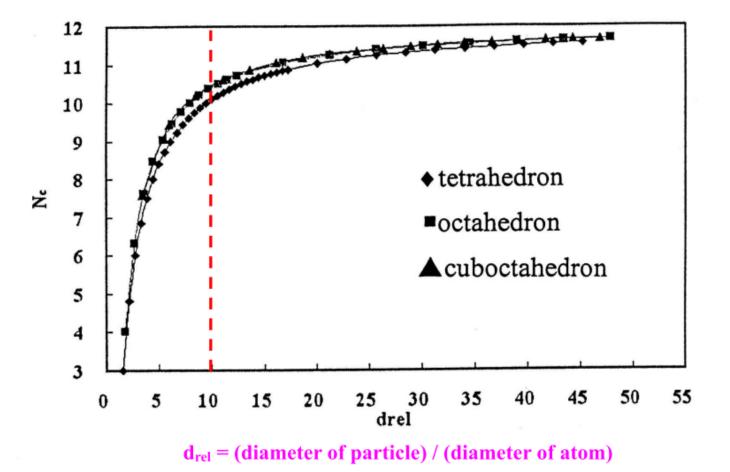


(a)Tetrahedron



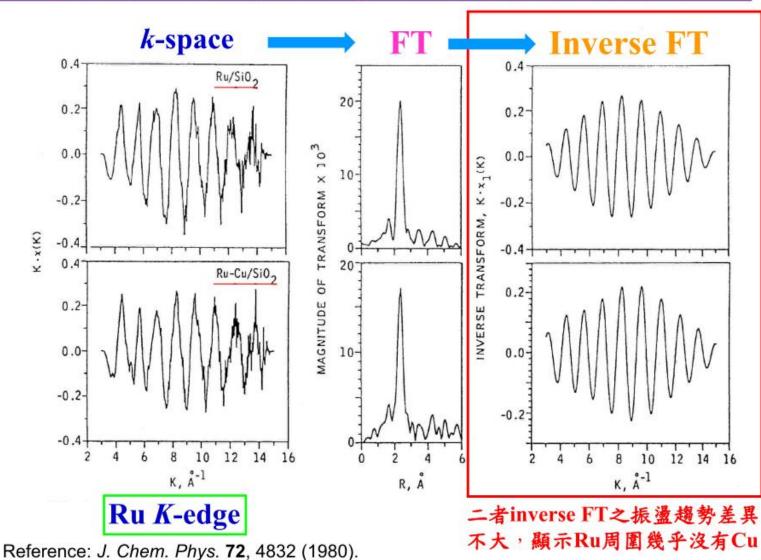


### Particle size越小 ⇒ 平均的配位數(N<sub>c</sub>)越低



三種理想 fcc 堆積模型之 Ne vs drel 關係圖

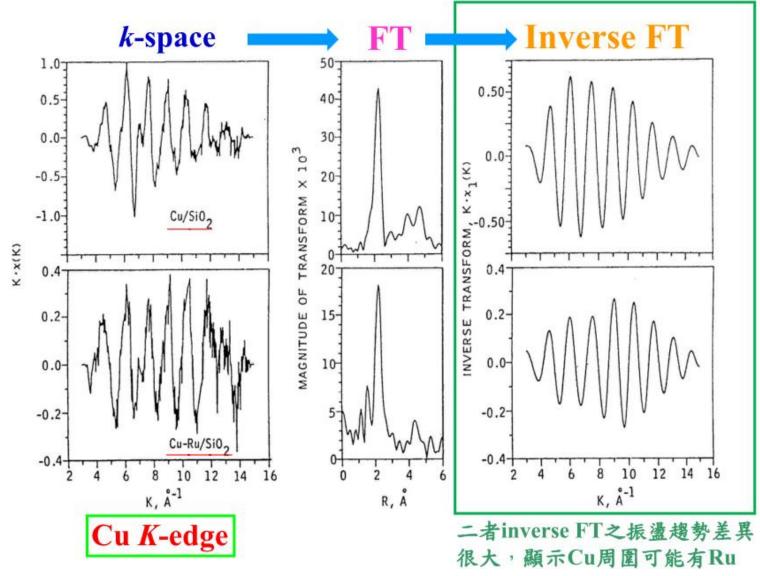




### **EXAFS to probe neighboring atoms of Ru-Cu nanoparticles**

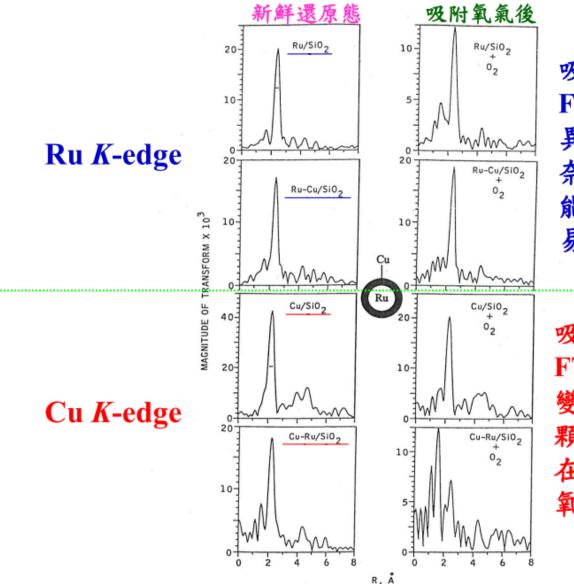








### **EXAFS to probe neighboring atoms of Ru-Cu nanoparticles**



吸附氧氣前後,Ru之 FT-EXAFS圖強度差 異不大;顯示Ru-Cu 奈米顆粒中的Ru可 能存在核心,因此不 易被氧化

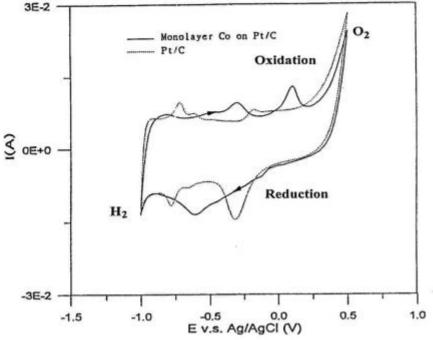
吸附氧氣之後,Cu之 FT-EXAFS圖明顯改 變;顯示Ru-Cu奈米 顆粒中的Cu可能存 在表面,因此容易被 氧化



## **Transformation of Co Monolayer on Pt/C**

- Sample was prepared via under-potential deposition (UPD).
- The Co monolayer can modify the reaction mechanism and hence the sensitivity of oxygen sensors.

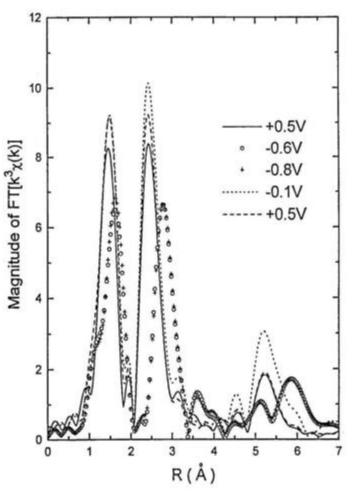
Cyclic voltammograms of Pt/C electrodes with and without monolayer Co in 1.0M KOH solution.





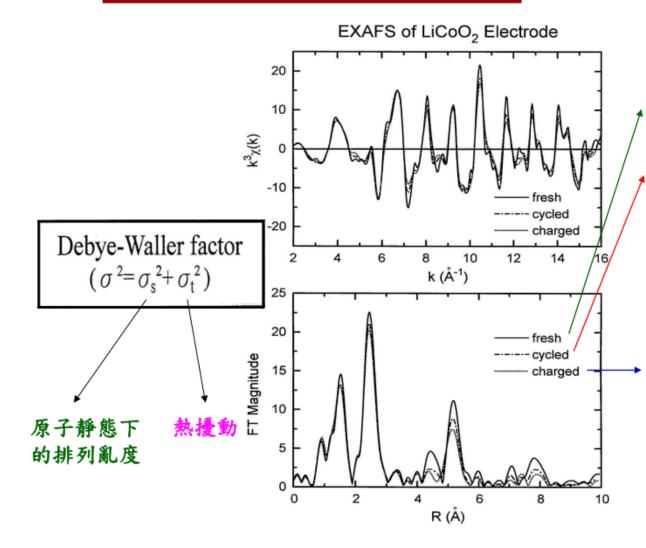
## Transformation of Co Monolayer on Pt/C

- Both XANES and EXAFS data indicated that Co is present as Co<sup>2+</sup> species when the applied voltage < -0.5 V.</li>
- Co is present as Co<sup>3+</sup> when the applied voltage > -0.2 V.
- The changes in electronic and structural properties are reversible.
- If the monolayer is Cu, the transformation will occur between Cu<sup>0</sup> and Cu<sup>2+</sup>.





#### EXAFS to probe structural disorder of LiCoO<sub>2</sub> electrode



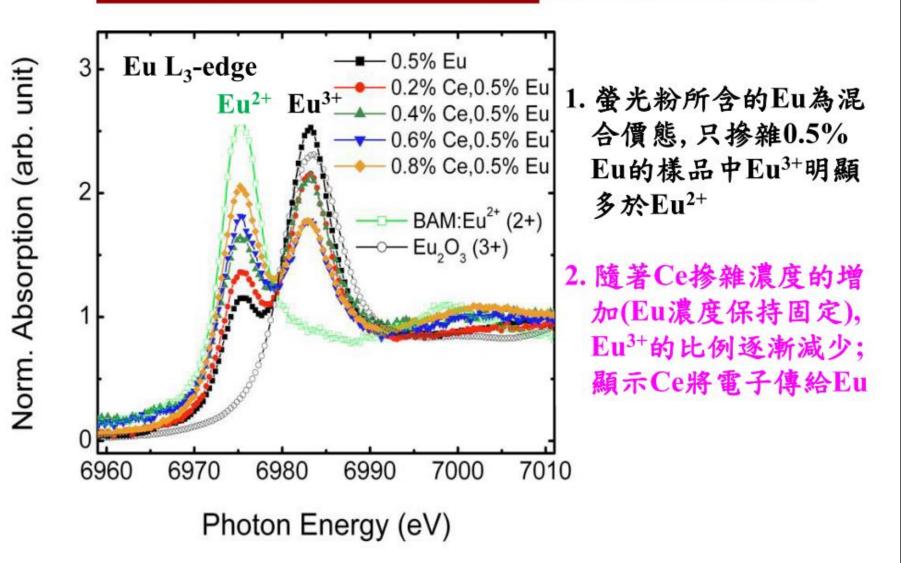
Fresh (初始態): 結構最有序

Cycled(充放電一次): 從圖上來看只能<u>部分</u> <u>或接近</u>回到初始狀態, 無法完全回復;因此手 機電池充放電循環多 次後,將使結構整個 破壞,無法繼續使用

Charged (充電): Li 離子從晶格遷出, 造成晶格部分扭曲, 故結構趨於disorder



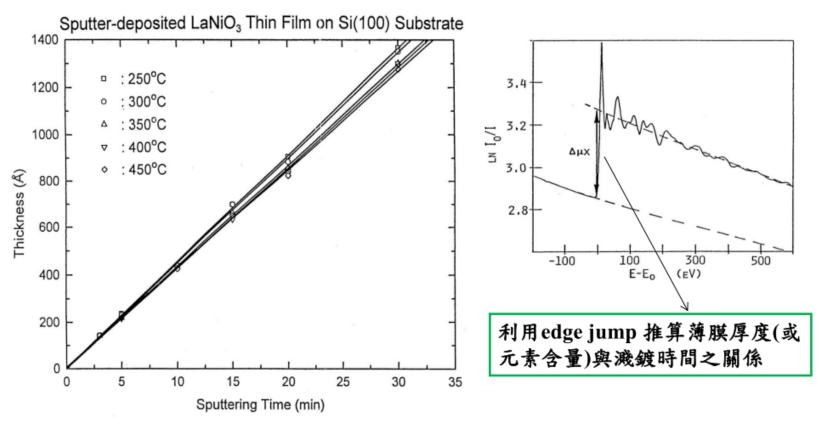
XANES to probe charge transfer of Li<sub>2</sub>SrSiO<sub>4</sub>:(Ce/Eu) yellow phosphor



Reference: Appl. Phys. Lett. 96, 061904 (2010).



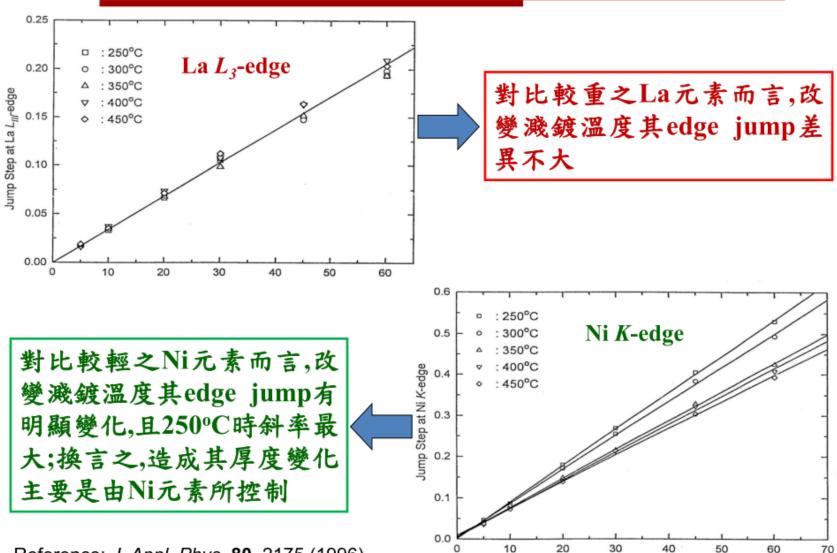
#### XANES to probe thickness of LaNiO<sub>3</sub> thin film



1. 改變不同Si(100)基板之濺鍍溫度;其薄膜厚度與濺鍍時間均呈正比

2. 基板之濺鍍溫度越低,其薄膜厚度越大(成長速度越快)





**XANES to probe thickness of LaNiO<sub>3</sub> thin film** 

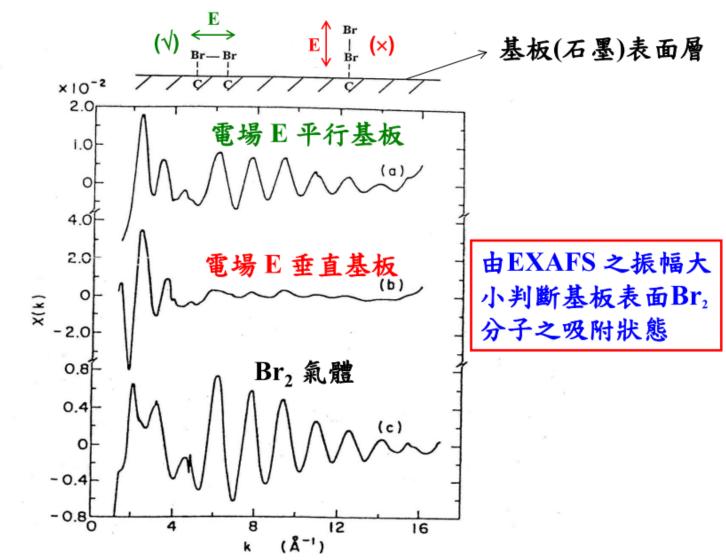
Reference: J. Appl. Phys. 80, 2175 (1996).



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Sputtering Time (min)

#### Polarized XAS investigation of Br<sub>2</sub> adsorbed on graphoil sample





# **Advantage of X-ray Absorption Spectroscopy**

- A powerful structural tool for materials in various forms, including crystalline or amorphous solids, liquids, and gases. Sample is not required to have a structure of long-range order.
- XANES reflects effective charge density (oxidation state), electronic structure, and coordination symmetry of the absorber.
- ➤ EXAFS provides the information of local (< 10 Å) atomic structure.</p>
- Element specific (element selective).
- Easy to perform *in-situ/operando* measurements.





獲得的訊息(包括氧化價數與配位數等) 為樣品中所有吸收原子的平均值,因 此當樣品含有不只一種狀態下的吸收 原子(譬如存在不同晶相中)時,解讀數 據分析結果需特別注意。



Thanks for your attentions